

SALT ANALYSIS

QUALITATIVE ANALYSIS OF INORGANIC SALT

Experiment Number,

Date:

AIM:

PRELIMINARY EXAMINATION OF SAMPLE No:			
1	Colour of the sample is noted	The sample is:	Presence of:
		Blue	Cu ²⁺ ions
		Green	Cu ²⁺ or Ni ²⁺ ions
		Dark yellow/Brown	Fe ³⁺ or Mn(NO ₃) ₂
		Flesh or light pink	Mn ²⁺ ions
2	A <u>thick paste</u> of the sample is prepared in con. HCl and is shown to the non-luminous part of a Bunsen flame with the help of a glass rod.	Colourless / White	Absence of transition metal cations.
		Livid Blue flame	Presence of Pb ²⁺ ion
		Green-edged blue flame	Presence of Cu ²⁺ ions.
		Crimson flame	Presence of Sr ²⁺ ions
		Pale green flame	Presence of Ba ²⁺ ions
		Brick red flame	Presence of Ca ²⁺ ion

ANALYSIS OF ANIONS

3	A little of the sample is treated with dil. HCl acid	Colourless, odourless gas (CO ₂) is evolved with effervescence.	Presence of CO ₃ ²⁻ ion
4	A little of the sample is treated with con. H ₂ SO ₄ .	White fuming, pungent smelling gas (HCl) produces dense white fumes when mouth of NH ₄ OH bottle is shown near the mouth of the test tube. Note: Warm/Heat if necessary	Presence of Cl ⁻ ion
5	A pinch of salt is heated with a few drops of conc. H ₂ SO ₄ for 10-15 seconds.	Reddish brown gas (NO ₂) on <u>boiling</u> the mixture with copper turnings and a few drops of conc. H ₂ SO ₄	Presence of NO ₃ ⁻ ion.
CONFIRMATORY TESTS FOR ANIONS			
6	An aqueous solution of the sample is acidified with dil.HCl followed by the addition of BaCl ₂ solution.	A white precipitate (BaSO ₄) formed, which is insoluble in con.HCl	Presence of SO ₄ ²⁻ ion
7	An aqueous solution of the sample is acidified with dilute acetic acid followed by the addition of lead acetate [(CH ₃ -	A white precipitate (PbSO ₄) formed, which is soluble in Ammonium acetate [CH ₃ COONH ₄] solution.	SO ₄ ²⁻ ion is confirmed.

Cbse Salt Analysis Manual

Henry M. Noad



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Practical/Laboratory Manual Chemistry Class - XI Er. Meera Goyal, 2021-05-29

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treatment method Viva Voce 3 Water Analysis 1 To test the hardness of different water samples Viva Voce 4 Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce

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Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 To purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method C Water Analysis 1 To test the hardness of different water samples D Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap E Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper F Analysis of Fruits and Vegetable Juices 1 To analyse the fruit and vegetable juices for the constituent present in them G Rate of Evaporation 1 To study the rate of evaporation of different liquids H Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Log Antilog Table

Comprehensive Practical Chemistry XII Dr. N . K. Verma, B. K. Vermani, Dr. Neera Verma, K . K. Rehani, 2011-11

Salt Analysis Chart Sibaji SARKAR, 2017-05-27 Salt analysis chart the present book is an attempt and an out come of the author experience of teaching the practical of inorganic chemistry subject for more than ten years which will immensely benefit the college and school students of the subject Primary aim to formulate this book is to reflect comprehensively the various aspect of inorganic chemistry and to acquaint students with sound knowledge of perform practical in inorganic chemistry which will be useful to them during the regular internal as external exam The book cover the basic of performing salt analysis in a new and exciting way a way that make it easier for students to learn and a way will be more interesting for teachers to teach I hope that both the students and teachers of science stream will receive this book favoring as a effective text book I sincerely hope that the book will go a long way to satisfy the long felt of students for a friendly book an practical in salt analysis I would highly appreciate receiving any input from users which might be of valuable of students at large The author will warmly welcome suggestion from reader at this book and necessary arrangement would be implemented to upgrade quality of book

Chemical Student's Manual for the Lecture Room and Laboratory. Embracing non-metals; the Analysis of Simple Salts, Etc Herbert L. BUCKERIDGE, 1887

LABORATORY MANUAL FOR SALT MIXING TEST IN ROD BUNDLES. , 1978

Handbook of Water Analysis Leo M.L. Nollet, Leen S. P. De Gelder, 2013-07-29 Extensively revised and updated Handbook of Water Analysis Third Edition provides current analytical techniques for detecting various compounds in water samples Maintaining the detailed and accessible style of the previous editions this third edition demonstrates water sampling and preservation methods by enumerating different ways to measure

c **Complex Salts** William Thomas, 1924

Manual of Chemical Analysis ... C. Greville Williams, 1858

Handbook of Pharmaceutical Salts Properties, Selection, and Use P. Heinrich Stahl, 2008-08-04 This comprehensive up to date guide and information source is an instructive companion for all scientists involved in research and development of drugs and in particular of pharmaceutical dosage forms The editors have taken care to address every conceivable aspect of the

preparation of pharmaceutical salts and present the necessary theoretical foundations as well as a wealth of detailed practical experience in the choice of pharmaceutically active salts Altogether the contributions reflect the multidisciplinary nature of the science involved in selection of suitable salt forms for new drug products *Single-salt Analysis* Brian Piers Lascelles,1896 **Laboratory Manual for Salt-mixing Test in 37- and 217-pin Bundles. [LMFBR].** ,1980 This laboratory manual deals with the procedure employed during salt tracer experiments used in evaluating the hydraulic characteristics of a rod bundle A description of the standard equipment used is given together with the details of manufacture of probes used for detecting the salt concentration Details of the bundle construction have been excluded as they are available in the reference cited An attempt has been made to point out potential trouble areas and procedures

Laboratory Manual for the Course in Advanced Quantitative Analysis Hobart Hurd Willard,1920 Muter's Short Manual of Analytical Chemistry John Muter,1927 Lab Manual Chemistry Class XII -by Dr. K. N. Sharma, Dr. Subhash Chandra Rastogi, Er. Meera Goyal (SBPD Publications) Dr. K. N. Sharma,Dr. Subhash Chandra Rastogi,Er. Meera Goyal,2021-07-03 Highly Useful for Various Engineering and Medical Competitive Examinations **Methods of Chemical Analysis Manual** Yorkshire Water Authority,1979 **The Salt Print Manual** Eleanor Young,2008-02-01 **A Manual of Chemical Analysis** Henry M. Noad,1863

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