

SALT ANALYSIS

QUALITATIVE ANALYSIS OF INORGANIC SALT

Experiment Number. -----

Date: -----

AIM:

PRELIMINARY EXAMINATION OF SAMPLE No:			
1	Colour of the sample is noted	The sample is:	Presence of:
		Blue	Cu^{2+} ions
		Green	Cu^{2+} or Ni^{2+} ions
		Dark yellow/Brown	Fe^{2+} or Mn^{2+} ions
		Flesh or light pink	Mn^{2+} ions
2	A <u>thick paste</u> of the sample is prepared in conc. HCl and is shown to the non-luminous part of a Bunsen flame with the help of a glass rod.	Colourless / White	Absence of transition metal cations.
		Livid Blue flame	Presence of Pb^{2+} ion
		Green-edged blue flame	Presence of Cu^{2+} ions
		Crimson flame	Presence of Sr^{2+} ions
		Pale green flame	Presence of Ba^{2+} ions
		Brick red flame	Presence of Ca^{2+} ion

ANALYSIS OF ANIONS

3	A little of the sample is treated with dil. HCl acid	Colourless, odourless gas (CO_2) is evolved with effervescence.	Presence of CO_3^{2-} ion
4	A little of the sample is treated with conc. H_2SO_4 .	White fuming, pungent smelling gas (HCl) produces dense white fumes when mouth of NH_4OH bottle is shown near the mouth of the test tube. <u>Note: Warm/Heat if necessary</u>	Presence of Cl^- ion
5	A pinch of salt is heated with a few drops of conc. H_2SO_4 for 10-15 seconds.	Reddish brown gas (NO_2) on <u>boiling</u> the mixture with copper turnings and a few drops of conc. H_2SO_4 .	Presence of NO_3^- ion.

CONFIRMATORY TESTS FOR ANIONS

6	An aqueous solution of the sample is acidified with dil. HCl followed by the addition of BaCl_2 solution.	A white precipitate (BaSO_4) formed, which is insoluble in conc. HCl	Presence of SO_4^{2-} ion
7	An aqueous solution of the sample is acidified with dilute acetic acid followed by the addition of lead acetate [$\text{CH}_3\text{COONH}_4$] solution.	A white precipitate (PbSO_4) formed, which is soluble in Ammonium acetate [$\text{CH}_3\text{COONH}_4$] solution.	SO_4^{2-} ion is confirmed.

Cbse Salt Analysis Manual

Henry M. Noad

Cbse Salt Analysis Manual:

CBSE Class 12th Chemistry (10 Most Likely Question Papers with Solution) By Career Point Kota Career Point Kota, 2020-07-28 The CBSE has made certain changes in the assessment structure from the session 2019 20 onwards In the new scheme of examination CCE and term system has been replaced with the Internal Assessment Single Annual Exam by CBSE itself Single exam conducted by CBSE will carry 80 marks whereas 20 marks are left to the schools for internal assessment CBSE has issued detailed guidelines on how the internal marks will be divided among different activities From 2019 onwards there will be internal choices in board examinations with increased internal options in the question paper Considering this change now a student has to prepare accordingly for board examinations The new assessment format brought with it excitement as well as anxiety And to help them prepare and excel in their CBSE board examination Career Point Kota has developed a series of 10 Most Likely Question Papers with Solutions The Key Features of Most Likely Question Papers with Solutions Series New OBJECTIVE TYPE question in each paper Syllabus of CBSE 2019 20 Based on the latest CBSE Syllabus Pattern Mind map of each chapter is given to visualize and help acquire a better understanding Important terms facts formulae and quick revision tips are given Covers questions asked in previous year board exams Toppers Answer Sheet as released by CBSE to understand the scoring technique We hope this book will gratify students need for the new CBSE pattern board exam and smoothen their path to success We wish to utilize the opportunity to place on record our special thanks to all the members of the Content Development team for their efforts to create this wonderful book

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1 Basic Laboratory Techniques 1 To cut a glass tube or glass rod 2 To bend the glass rod at an angle 3 To draw a glass jet from a glass tube 4 To bore a cork and fit a glass tube into it Viva Voce 2 Characterisation and Purification of Chemical Substances 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique Viva Voce 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method Viva Voce 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample 4 To prepare the pure crystals of copper sulphate from the given crude sample 5 To prepare pure crystals of benzoic acid from a given impure sample Viva Voce 3 Measurement of pH Values 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+ Viva Voce 4 Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2 To study the shift in equilibrium between Co^{2+} and Cl^- ions by changing the concentrations of either of the ions Viva Voce 5 Quantitative Analysis 1 To prepare M 10 oxalic acid solution by direct weighing method 2 To prepare M 10 solution of sodium carbonate by direct weighing method 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution Viva Voce 6 Qualitative Analysis Analysis of Anions Analysis of Cations Viva Voce 7 Detection of Elements in Organic Compounds 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test Viva Voce INVESTIGATORY PROJECTS 1 Checking of Bacterial Contamination in Water 1 To check the bacterial contamination in drinking water by testing sulphide ions Viva Voce 2 Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 To purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone

treatment method Viva Voce 3 Water Analysis 1 To test the hardness of different water samples Viva Voce 4 Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce

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Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 To purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method C Water Analysis 1 To test the hardness of different water samples D Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap E Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper F Analysis of Fruits and Vegetable Juices 1 To analyse the fruit and vegetable juices for the constituent present in them G Rate of Evaporation 1 To study the rate of evaporation of different liquids H Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Log Antilog Table *Comprehensive Practical Chemistry XII* Dr. N . K. Verma,B. K. Vermani,Dr. Neera Verma,K . K. Rehani,2011-11

Salt Analysis Chart Sibaji SARKAR,2017-05-27 Salt analysis chart the present book is an attempt and an out come of the author experience of teaching the practical of inorganic chemistry subject for more than ten years which will immensely benefit the college and school students of the subject Primary aim to formulate this book is to reflect comprehensively the various aspect of inorganic chemistry and to acquaint students with sound knowledge of perform practical in inorganic chemistry which will be useful to them during the regular internal as external exem The book cover the basic of performing salt analysis in a new and exciting way a way that make it easier for students to learn and a way will be more interesting for teachers to each I hope that both the students and teachers of science stream will receive this book favoring as a effective text book I sincerely hope that the book will go a long way to satisfy the long felt of students for a friendly book an practical in salt analysis I would highly appreciate receiving any input from users which might be of valuable of students at large The author will warmly welcome suggestion from reader at this book and necessary arrangement would be implemented to upgrade quality of book Chemical Student's Manual for the Lecture Room and Laboratory. Embracing-non-metals; the Analysis of Simple Salts, Etc Herbert L. BUCKERIDGE,1887

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MANUAL FOR SALT MIXING TEST IN ROD BUNDLES. ,1978 Handbook of Water Analysis Leo M.L. Nollet,Leen S. P. De Gelder,2013-07-29 Extensively revised and updated Handbook of Water Analysis Third Edition provides current analytical techniques for detecting various compounds in water samples Maintaining the detailed and accessible style of the previous editions this third edition demonstrates water sampling and preservation methods by enumerating different ways to measure c **Complex Salts** William Thomas,1924 **Manual of Chemical Analysis ...** C. Greville Williams,1858 Handbook of Pharmaceutical Salts Properties, Selection, and Use P. Heinrich Stahl,2008-08-04 This comprehensive up to date guide and information source is an instructive companion for all scientists involved in research and development of drugs and in particular of pharmaceutical dosage forms The editors have taken care to address every conceivable aspect of the

preparation of pharmaceutical salts and present the necessary theoretical foundations as well as a wealth of detailed practical experience in the choice of pharmaceutically active salts Altogether the contributions reflect the multidisciplinary nature of the science involved in selection of suitable salt forms for new drug products *Single-salt Analysis* Brian Piers Lascelles,1896 **Laboratory Manual for Salt-mixing Test in 37- and 217-pin Bundles. [LMFBR].** ,1980 This laboratory manual deals with the procedure employed during salt tracer experiments used in evaluating the hydraulic characteristics of a rod bundle A description of the standard equipment used is given together with the details of manufacture of probes used for detecting the salt concentration Details of the bundle construction have been excluded as they are available in the reference cited An attempt has been made to point out potential trouble areas and procedures

Laboratory Manual for the Course in Advanced Quantitative Analysis Hobart Hurd Willard,1920 Muter's Short Manual of Analytical Chemistry John Muter,1927 Lab Manual Chemistry Class XII -by Dr. K. N. Sharma, Dr. Subhash Chandra Rastogi, Er. Meera Goyal (SBPD Publications) Dr. K. N. Sharma,Dr. Subhash Chandra Rastogi,Er. Meera Goyal,2021-07-03 Highly Useful for Various Engineering and Medical Competitive Examinations **Methods of Chemical Analysis Manual** Yorkshire Water Authority,1979 **The Salt Print Manual** Eleanor Young,2008-02-01 **A Manual of Chemical Analysis** Henry M. Noad,1863

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