

Chapter 8 - Chemical Equations and Reactions

8-1 Describing Chemical Reactions

I. Introduction

A. Reactants

1. Original substances entering into a chemical rxn

B. Products

1. The resulting substances from a chemical rxn



C. Chemical Equation

1. Represents with symbols and formulas, the identities and relative amounts of the reactants and products in a chemical rxn

II. Indications of a Chemical Reaction

A. Evolution of Heat and Light

1. Evidence of energy being released (exothermic rxn)

B. Production of a Gas

1. CO_2 , H_2 , H_2S are some gases produced by chemical rxns



C. Formation of a Precipitate

1. Precipitate is a solid that is produced as a result of a chemical rxn in solution



III. Characteristics of Chemical Equations

A. The equation must represent known facts

1. This can be done with a word equation

"hydrogen reacts with oxygen to form water"



B. The equation must contain the correct formulas for reactants and products

1. This is done with a formula equation



C. The law of conservation of atoms must be satisfied

1. Balancing is done with coefficients - small whole numbers that appear in front of a formula



D. Additional symbols used in Chemical equations

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Sebastian Brünink



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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03

The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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Modern Inorganic Chemistry J. J. Lagowski, 1973 **American Magazine**, 1913 **Industrial & Engineering Chemistry**, 1926 *The American Magazine*, 1913 *American Illustrated Magazine*, 1913 **Technical World Magazine**, 1913 **Building Age**, 1928

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