

Concentration and Molarity (M) table

Name: _____

Mass concentration (density)

Using the concentration and the formula for density, calculate the density

Mass of solute (g)	Mass of solvent (g)	Mass of solution (g)	Mass of solute (g)	Mass of solvent (g)	Mass of solution (g)
10	70	80.7	10.000	70	80
20	80	91.5	20	70.5	90.5
3.0	70	73.0	3.0	67.0	70.0
20	20	40.0	54.4	6.4	60.8

Concentration (density and molarity) concentration and Molarity (M) table

1. **What is the mass of a 100 mL solution with a density of 1.05 g/mL?**
 $100 \text{ mL} \times 1.05 \text{ g/mL} = 105 \text{ g}$
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Answer Key

B Lingard



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