

A close-up photograph of a person's hands. The left hand is holding a dark, cylindrical pill bottle, tilted to pour several white, oval-shaped pills into the palm of the right hand. The background is blurred, showing what appears to be a clinical or laboratory setting with various pieces of equipment.

7 SECRETS TO LONGEVITY

From Dr. David Sinclair

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Longevity Secrets Quick Start

Lingsheng Yao



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web solution $v_1 = 250 \text{ l}$ $v_2 = 300 \text{ l}$ $t_1 = 273 \text{ K}$ $t_2 = 300 \text{ K}$ we will be applying charles law as pressure is constant here $v_1/t_1 = v_2/t_2$ or $v_2 = v_1 \cdot t_2 / t_1 = 250 \text{ l} \cdot 300 \text{ K} / 273 \text{ K} = 278 \text{ l}$ at 90°C a helium sample has a volume of 500 ml determine the temperature at which the volume of the liquid will become 240 ml