

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write yes if evidence of a chemical reaction is present. Write no if there is no evidence of a chemical reaction.

yes 1. A tomato smells rotten.

no 2. A drinking glass breaks into smaller pieces.

no 3. A piece of ice melts.

yes 4. Drain cleaner is mixed with water and the solution becomes warm.

yes 5. Candle wax burns.

yes/no 6. Molten candle wax solidifies.

yes 7. Green leaves turn yellow and red as the seasons change.

yes 8. Baking powder produces a gas that makes a cake rise.

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(g)	(g)
reactant	product	(g)	(g)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is also (9) reactant, oxygen is also (9) reactant, and water vapor is also (9) product. In a chemical equation for this reaction, also (12) arrow is used to separate hydrogen and oxygen from water vapor and energy. Also (13) plus sign is used to separate the symbols for hydrogen and oxygen. Also (14) (g) symbol is used to tell the state of hydrogen in the reaction, also (15) (g) symbol is used for the state of oxygen, and also (16) (g) symbol is used for the state of water vapor.

Chemistry Study Guide Key

R Sandford

Chemistry Study Guide Key:

Organic Chemistry Study Guide Robert J. Ouellette,J. David Rawn,2014-10-13 Organic Chemistry Study Guide Key Concepts Problems and Solutions features hundreds of problems from the companion book Organic Chemistry and includes solutions for every problem Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject Organic chemistry is a constantly evolving field that has great relevance for all scientists not just chemists For chemical engineers understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant For biologists and health professionals it is essential because nearly all of biochemistry springs from organic chemistry Additionally all scientists can benefit from improved critical thinking and problem solving skills that are developed from the study of organic chemistry Organic chemistry like any skill is best learned by doing It is difficult to learn by rote memorization and true understanding comes only from concentrated reading and working as many problems as possible In fact problem sets are the best way to ensure that concepts are not only well understood but can also be applied to real world problems in the work place [Study Guide, Answer Book to](#)

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