

Concentration Worksheet – Answers

- 1) How many grams of beryllium chloride are needed to make 125 mL of a 0.050 *M* solution?
0.50 grams
- 2) How many grams of beryllium chloride would you need to add to 125 mL of water to make a 0.050 molal solution?
0.50 grams
- 3) The density of ethanol is 0.789 g/mL. How many grams of ethanol should be mixed with 225 mL of water to make a 4.5% (v/v) mixture?
7.99 grams
- 4) Explain how to make at least one liter of a 1.25 molal ammonium hydroxide solution.
Dissolve 43.8 grams (1.25 moles) of ammonium hydroxide in 1 L H₂O.
- 5) What is the molarity of a solution in which 0.45 grams of sodium nitrate are dissolved in 265 mL of solution.
0.020 *M*
- 6) What is the mole fraction of sulfuric acid in a solution made by adding 3.4 grams of sulfuric acid to 3,500 mL of water?
1.8 × 10⁻⁴
- 7) What will the volume of a 0.50 *M* solution be if it contains 25 grams of calcium hydroxide?
680 mL
- 8) How many grams of ammonia are present in 5.0 L of a 0.050 *M* solution?
4.3 grams

Chemistry Study Guide Solution Concentration Answers

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Organic Chemistry Study Guide and Solutions Marc Loudon, Jim Parise, 2015-07-01 Parise and Loudon's Study Guide and Solutions Manual offers the following learning aids: Links that provide hints for study approaches to problem solving and additional explanations of challenging topics; Further Explorations that provide additional depth on key topics; Reaction summaries that delve into key mechanisms and stereochemistry; Solutions to all the textbook problems. Rather than providing just the answer, many of the solutions provide detailed explanations of how the problem should be approached.

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Introduction to Chemistry, Study Guide T. R. Dickson, 1999-07-07 Teaches chemistry by offering a dynamic, provocative, and relevant view of the topic and its importance to society and our daily lives. Three themes are stressed throughout the text: developing chemical thinking, chemical vision, and refining problem-solving skills. Many chapters in this edition have been rewritten and rearranged to vitalize the topics and to include interesting examples, analogies, and images.

Student Study Guide/solutions Manual to Accompany Foundations of Inorganic, Organic, and Biological Chemistry Katherine J. Denniston, 1995 This student study guide has been written to accompany *Foundations of Inorganic, Organic, and Biological Chemistry* by Caret Denniston and Topping. It was designed to complement the text, not to be used in place of the text. Each chapter of the study guide contains the following sections: reorganized and expanded set of learning goals, concise chapter summary, in chapter solved problems, list of key terms, review problems, preface.

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Study Guide to Accompany Introduction to Chemistry J.C. Cummins,Jack D. Cummins,1975 Fundamentals of Chemistry, Study Guide James E. Brady,John R. Holum,1988-04-20 This Third Edition of the widely used fundamentals textbook for science majors maintains the conversational writing style that made the previous editions so popular while including up to date treatments of important and current topics Emphasizes descriptive chemistry chemical reactions and properties while maintaining a solid treatment of chemical principles Common chemicals are used whenever possible as examples in both theoretical discussions and in problems and exercises Incorporates many pedagogical aids each chapter begins with a brief table of contents and each section begins with a preview of topics covered Chapters include frequent margin comments figures and photographs

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