

AP Chemistry Final Study Guide

Unit 1: Atomic Structure and Properties (7%-9%)

- At STP, one mole of gas always occupies 22.4 L
- **Mass Spectroscopy:**
 - Atoms have different isotopes (carbon-12, carbon-14, etc) and this will affect the average atomic mass
 - The most abundant isotope will have the greatest effect
- **Coulomb's Law:**
 - The greater the charge of the nucleus, the more energy is required to ionize an electron
 - The closer an electron is to the nucleus, the more energy is required to ionize an electron
- **Electromagnetic Radiation:**
 - When atoms absorb energy, electrons jump to higher energy levels
 - When electrons drop from higher to lower energy levels, atoms give off electromagnetic radiation energy
- **Kinetic Energy of Ionized Electrons:**
 - The faster an ejected electron is going, the more kinetic energy it has
 - Electrons originally further away from the nucleus require less energy to eject, so it will have more kinetic energy
- **Electron Configuration:**
 - When entering the d-block (transition metals), the energy level drops by 1 unless you are using shorthand notation
 - Nickel: $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 4s^2$ (notice the d-block)
 - Nickel (with shorthand): $[\text{Ar}]4s^2 3d^8$
- **Law of Definite Proportions:**
 - A pure chemical compound broken down into elements always contains elements of a fixed ratio, independent of where and how it was created
- **Law of Multiple Proportions:**
 - When elements form compounds, the proportions of the elements in those chemical compounds can be expressed in small whole number ratios
- **Aufbau Principle:**
 - When building up the electron configuration of an atom, electrons are placed in orbitals, subshells, and shells in order of increasing energy

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Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011 **Student Study Guide Fundamentals of Chemistry by David E. Goldberg** Margaret G. Kimble, 1994 **Robinson Chemistry Study Guide** Robinson, 1992

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day and beyond [Organic Chemistry](#) National Learning Corporation,2020-03-15 **Learning Science Through Computer Games and Simulations** National Research Council,Division of Behavioral and Social Sciences and Education,Board on Science Education,Committee on Science Learning: Computer Games, Simulations, and Education,2011-05-12 At a time when scientific and technological competence is vital to the nation's future the weak performance of U.S. students in science reflects the uneven quality of current science education. Although young children come to school with innate curiosity and intuitive ideas about the world around them science classes rarely tap this potential. Many experts have called for a new approach to science education based on recent and ongoing research on teaching and learning. In this approach simulations and games could play a significant role by addressing many goals and mechanisms for learning science: the motivation to learn science, conceptual understanding, science process skills, understanding of the nature of science, scientific discourse and argumentation, and identification with science and science learning. To explore this potential **Learning Science Computer Games Simulations and Education** reviews the available research on learning science through interaction with digital simulations and games. It considers the potential of digital games and simulations to contribute to learning science in schools, in informal out-of-school settings and everyday life. The book also identifies the areas in which more research and research-based development is needed to fully capitalize on this potential. **Learning Science** will guide academic researchers, developers, publishers and entrepreneurs from the digital simulation and gaming community and education practitioners and policy makers toward the formation of research and development partnerships that will facilitate rich intellectual collaboration. Industry, government agencies and foundations will play a significant role through start-up and ongoing support to ensure that digital games and simulations will not only excite and entertain but also motivate and educate.

ASAP Chemistry: a Quick-Review Study Guide for the AP Exam The Princeton Review,2018-11-15 Subtitle on cover A quick review study guide for the AP Exam [Directory of Distance Learning Opportunities](#) Modoc Press, Inc.,2003-02-28 This book provides an overview of current K-12 courses and programs offered in the United States as correspondence study or via such electronic delivery systems as satellite cable or the Internet. The Directory includes over 6,000 courses offered by 154 institutions or distance learning consortium members. Following an introduction that describes existing practices and delivery methods, the Directory offers three indexes: Subject Index, Index of Courses Offered by Level, Course Level Index, Geographic Index. All information was supplied by the institutions. Entries include current contact information, a description of the institution and the courses offered, grade level and admission information, tuition and fee information, enrollment periods, delivery information, equipment requirements, credit and grading information, library services and accreditation. [NES Chemistry Study Guide](#) NES Chemistry Exam Prep Team,2018-02-12 Imagine a study guide actually designed for teachers. Because we know you've got a busy life, we've developed a study guide that isn't like other certification materials out there. With our NEW NES Chemistry Study Guide Test Prep and Practice Questions for the National Evaluation Series Chemistry

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de kok 250 jaar van kokshuis tot boerderij 17502000

de lotgevallen van witkop in de stad illustr jshl

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de knuffels met verhalen van eigenaren van knuffels

de nederlandsche natie het nederlandsche volkskarakter

de levende gedachten van dostojewski belicht door rob limburg

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