

ChemQuest 1.1

Intro. to Chemical Reactions

Name: _____
 Date: _____
 Class: _____

Electrolyte Review

1. Write the formula for the following compounds:

- a) barium phosphate: $\text{Ba}_3(\text{PO}_4)_2$ b) sodium phosphate: Na_3PO_4
 c) magnesium phosphate: $\text{Mg}_3(\text{PO}_4)_2$ d) lithium nitrate: LiNO_3

2. Write the name for the following compounds:

- a) $\text{Cu}(\text{NO}_3)_2$: copper(II) nitrate b) FeCl_3 : iron(III) chloride
 c) SnSO_4 : tin(II) sulfate d) SnF_4 : tin(IV) fluoride

Chemical Reactions: What they look like...

Here is what a chemical reaction looks like when it is written:



There are examples that you can use to describe what happens during:

Combustion reactions react with oxygen and make oxides, and energy.

There are also a lot of other reactions that you can use to describe what happens during. Here are a few more:

Critical Thinking Questions

1. Write the following reaction, with the correct state symbols.



Balance the reaction with the correct state symbols and energy symbols.

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L Cohen



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