

Chemistry: Chapter 62 Separation and purification methods

Multiple-choice questions

Sections 62.1–62.2

|||CMEMF0156200001|||

Which of the following statements about crystallization are correct?

- (1) Crystals obtained by slow cooling are purer than those obtained quickly.
- (2) After filtration, we should use a large amount of distilled water to wash the crystals in order to remove any soluble impurities.
- (3) Crystallization is a physical method that separates a dissolved substance from a solution.

- A. (1) and (2) only
- B. (1) and (3) only
- C. (2) and (3) only
- D. (1), (2) and (3)



##B## We should use a little cold distilled water to wash the crystals after filtration, otherwise, the crystals may dissolve.##

|||CMEMF0156200002|||

Which of the following methods is the most suitable to obtain magnesium sulphate crystal from magnesium sulphate solution?

- A. Simple distillation
- B. Crystallization
- C. Fractional distillation
- D. Solvent extraction



##B##

Section 62.3

|||CMEMF0156200003|||

Which of the following substances CANNOT be separated by fractional distillation?

- (1) A mixture of proteins
- (2) Crude oil fractions
- (3) Oxygen and nitrogen in air

- A. (1) only
- B. (2) only

Chromatography Multiple Choice Questions

JL Elias

Chromatography Multiple Choice Questions:

TEXTBOOK OF INSTRUMENTAL METHODS OF ANALYSIS Dr. Santosh Karajgi, Dr. Archana Swamy P , Mrs. Anitha L, Mr. P. S. Minhas , Dr. Varinder Soni,2024-05-15 Introducing the book Instrumental Methods of Analysis is something that fills me with an incredible amount of joy The content of this book has been meticulously crafted to adhere to the curriculum for Bachelor of Pharmacy students that has been outlined by the Pharmacy Council of India An effort has been made to investigate the topic using terminology that is as straightforward as possible in order to make it more simply digestible for pupils The book has a number of illustrations such as flowcharts and diagrams that make it simple for students to comprehend complex ideas It is the author s honest desire that both students and academicians would take something helpful away from reading this book

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