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1.

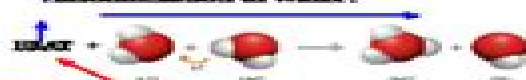
## Unit 8

### Acids & Bases

2.

### 8.1 Introduction to Acids & Bases

#### Autoionization of Water:



This is an endothermic process- increasing the heat will shift the reaction to produce more  $\text{H}_3\text{O}^+$  and  $\text{OH}^-$ .  $K_w$  increases (more products), both pH and pOH will decrease but pure H<sub>2</sub>O will still be neutral.

$$K_w = [\text{H}_3\text{O}^+][\text{OH}^-]$$

Only digits to the right of the decimal are significant in pH and pOH values. See

$$\text{pH} = 3.45$$

is sig. figs

In case of acidic solution  $> [\text{H}^+] > [\text{OH}^-]$

$$\text{pH} < 7.00 \text{ at } 25^\circ\text{C}$$

In case of neutral solution  $> [\text{H}^+] = [\text{OH}^-]$

$$\text{pH} = 7.00 \text{ at } 25^\circ\text{C}$$

In case of basic solution  $> [\text{H}^+] < [\text{OH}^-]$

$$\text{pH} > 7.00 \text{ at } 25^\circ\text{C}$$

#### ION PRODUCT CONSTANT:

$$K_w = [\text{H}^+][\text{OH}^-] = 1.0 \times 10^{-14} \text{ at } 25^\circ\text{C}$$

$$K_w = K_a \times K_b$$

$$\text{pH} = -\log[\text{H}^+], \text{pOH} = -\log[\text{OH}^-]$$

$$14 = \text{pH} + \text{pOH}$$

Also known:

$$[\text{H}^+] = 10^{-\text{pH}} \text{ and } [\text{OH}^-] = 10^{-\text{pOH}}$$

3. The value of  $K_w$  for water at  $0^\circ\text{C}$  is  $1 \times 10^{-15}$ . What is the pOH of water at  $0^\circ\text{C}$ ?

a) 7.5

b) 8.0

c) 7.0

d) 6.5

# Chemistry Unit 8 Review Answers

**SB Merriam**

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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**Chemistry** Royal B. Leach, Galen Wood Ewing, 1966 *Chemistry: Matter & Change, Solving Problems - A Chemistry*

*Handbook* McGraw Hill, 2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change

*Qualitative Analysis and Ionic Equilibrium* George H. Schenk, 1990 A supplement for courses with a qualitative analysis component this lab manual contains explanations of the chemistry of metal ions and anions It includes pre lab exercises experiments and lab reports *Pennsylvania Township News*, 1974 *McGraw-Hill Education 500 Review Questions for the MCAT: General Chemistry* Mary Millhollon, Richard H. Langley, 2015-07-10 500 ways to pass the General Chemistry section of the new MCAT Intensive practice detailed answer explanations the best way to sharpen skills and prepare for the exam In anticipation of the fully revised 2015 MCAT 500 Review Questions for the MCAT General Chemistry has been updated to comprehensively cover the chemistry portion of the Chemical and Physical Foundations of Biological Systems section This book provides the problem solving practice you need to take the exam with confidence 500 questions organized by subject Follows the new MCAT format Complete explanations to every question given in the answer key **Descriptive Chemistry** Lyman Churchill Newell, 1903 *Review of Nurse Anesthesia* John J. Nagelhout, 1999 Organized according to the certification examination for CRNAs this review book presents 1000 classroom tested questions accurate answers comprehensive rationales and references Plus it includes a 3 1 2 disk for Windows tm or Macintosh r which puts every question at users fingertips **Nursing School Entrance Exam**, 2005-11 Discusses career opportunities in nursing offers test taking strategies and includes three full length practice exams **Iron & Coal Trades Review**, 1922 Vol 115 includes Diamond jubilee issue 1867 1927 **Books in Print Supplement**, 2002

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