

Chapters 1 and 2 Study Guide

- Solve the following equation for P_1 : $P_1 P_2 = P_2 P_1 \Rightarrow \frac{P_1 P_2}{P_2} = \frac{P_2 P_1}{P_1} \Rightarrow P_1 = P_2$
- Solve the following equation for m : $m = m \times 10^3 \times 10^3 \times 10^3$
- Write each of the following in scientific notation:
 - 3500
 - 0.0000016
- Write each of the following in scientific notation:
 - 425 000
 - 0.00000018
- State the type of measurement indicated in each of the following:
 - 45.6 g
 - 1.85 L
 - 14 cm
 - 6.3×10^3 mg
- State the type of measurement indicated in each of the following:
 - 15 °C
 - 25 s
- Identify the significant zeros in each of the following measured numbers:
 - 0.000250 m
 - 70.040 g
 - 1.000000 L
- Identify the significant zeros in each of the following measured numbers:
 - 0.04000 m
 - 6.00×10^7 g
- Round off each of the following numbers to three significant figures:
 - 35.7823 m \Rightarrow 35.78 m
 - 0.002 621 7 L \Rightarrow 0.00262 L
 - 3.8268×10^7 g \Rightarrow 3.83 $\times 10^7$ g
- Perform the following calculations of measured numbers and give the answers with the correct number of significant figures:
 - 45.26×0.01038
 $\Rightarrow 45.26 \times 1.038 \times 10^{-2}$
 $\Rightarrow 4.624$
 - $2.4 \times 3.34 \times 0.00700$
 $\Rightarrow 2.4 \times 3.34 \times 7.00 \times 10^{-5}$
 - $\frac{4.0 \times 8.00}{16}$
 $\left(2 \text{ significant digits} \right)$

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Alan S. Wingrove, Robert L. Caret

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Solomons,Craig B. Fryhle,Scott A. Snyder,2016-04-11 This is the Student Study Guide Solutions Manual to accompany

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Study Guide with Answers to Selected Problems H. Stephen Stoker, Danny V. White, 2003-04. This supplement includes for each chapter a brief overview, activities and practice problems to reinforce skills and a practice test. The answers section includes answers for all odd numbered end of chapter exercises. Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition Susan McMurry, 2000. Provides answers and explanations to all in text and end of chapter exercises. Also includes summaries of name reactions, functional group synthesis and reactions, lists of reagents and abbreviations, and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry. This edition now includes all new artwork.

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