

So the mean kinetic energy of a nitrogen molecule is the same as the mean kinetic energy of an oxygen molecule since E_k is directly proportional to temperature $E_k = \frac{3}{2}kT$

b) molar mass $M_n = 0.028 \text{ kg mol}^{-1}$ molar mass $M_o = 0.032 \text{ kg mol}^{-1}$
 $\frac{M_n}{M_o} = \frac{6.02 \times 10^{23}}{6.02 \times 10^{23}}$

$$\frac{M_n}{M_o} \text{ cm}_n^2 = \frac{0.028}{0.032} \text{ cm}_o^2$$

$$\frac{0.028}{6.02 \times 10^{23}} \text{ cm}_n^2 = 0.028 \text{ cm}_o^2$$

$$\text{cm}_o^2 = \frac{0.028}{0.028} \text{ cm}_n^2$$

$$\text{cm}_o^2 = 1.07 \text{ cm}_n^2$$

$$\text{cm}_o = \sqrt{1.07} \text{ cm}_n$$

$$\text{cm}_o = 1.07 \text{ cm}_n$$

4. molar mass = 0.028 kg/mol $V = 0.077 \text{ m}^3$ $T = 300 \text{ K}$
 $p = 100 \text{ kPa} = 100 \times 10^3 \text{ Pa}$

a) $pV = nRT \Rightarrow \frac{pV}{RT} = n = \frac{(100 \times 10^3)}{8.31 (300)} (0.028)$
 $= \frac{3700}{2493} = 1.484 = 1.484 \text{ mole to 3SF}$

b) mass = number of mole \times molar mass
 $= 1.484 \times 0.028 = 0.04163 \text{ kg}$
 $= 4.16 \times 10^{-2} \text{ kg to 2SF}$

If 1.48 mole used then mass = $0.04163 \text{ kg} + 4.1 \times 10^{-2} \text{ kg to 2SF}$

c) mass of one molecule $\text{or a molecule} = \frac{0.028}{6.02 \times 10^{23}} = 4.67 \times 10^{-26} \text{ kg}$

$$\frac{1}{2} m \text{ cm}_n^2 = \frac{3}{2} kT$$

$$4.67 \times 10^{-26} \text{ cm}_n^2 = \frac{3}{2} (1.38 \times 10^{-23}) 300 = 1.242 \times 10^{-24}$$

$$\text{cm}_n^2 = \frac{1.242 \times 10^{-24}}{4.67 \times 10^{-26}} = 26.7 \times 10^2 \text{ cm}_n^2$$

$$\text{cm}_n^2 = 26.7 \times 10^2$$

$$= 5.16 \text{ m}^2$$

$$= 5.2 \times 10^{-2} \text{ m}^2 \text{ to 2SF}$$

W.C calculate
 $\frac{1}{2} m \text{ cm}_n^2$

Some time mass
 and molar mass
 equals to $1.48 (0.028)$

Conceptual Physics Chapter 20 Gases Exercises Answers

Raymond A. Serway

Conceptual Physics Chapter 20 Gases Exercises Answers:

Study Guide for Physics in the Modern World 2E Jerry Marion, 2012-12-02 Study Guide for Physics in the Modern World 2E provides information pertinent to the fundamental concepts in physics. This book presents a list of concepts, definitions, and equations with various supplementary exercises for the readers. Comprised of 21 chapters, this book starts with an overview of the standard units of measure for length, time, mass, energy, force, pressure, and density. This text then provides the meaning of various terms in physics, including atom, molecule, element, and compound. Other chapters explore the composition and behavior of all ordinary matter in which it depends on the four basic units, including electrons, protons, neutrons, and photons. This book discusses as well the method used for converting the units of physical quantities from one system of measurement to another. The final chapter deals with the various applications of radiation in biological investigations as well as in medical diagnostics and therapeutics. This book is intended for students enrolled in introductory physics courses.

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closer to the way physics is actually practiced Key Topics INTRODUCTION MEASUREMENT ESTIMATING DESCRIBING MOTION KINEMATICS IN ONE DIMENSION KINEMATICS IN TWO OR THREE DIMENSIONS VECTORS DYNAMICS NEWTON S LAWS OF MOTION USING NEWTON S LAWS FRICTION CIRCULAR MOTION DRAG FORCES GRAVITATION AND NEWTON S6 SYNTHESIS WORK AND ENERGY CONSERVATION OF ENERGY LINEAR MOMENTUM ROTATIONAL MOTION ANGULAR MOMENTUM GENERAL ROTATION STATIC EQUILIBRIUM ELASTICITY AND FRACTURE FLUIDS OSCILLATIONS WAVE MOTION SOUND TEMPERATURE THERMAL EXPANSION AND THE IDEAL GAS LAW KINETIC THEORY OF GASES HEAT AND THE FIRST LAW OF THERMODYNAMICS SECOND LAW OF THERMODYNAMICS ELECTRIC CHARGE AND ELECTRIC FIELD GAUSS S LAW ELECTRIC POTENTIAL CAPACITANCE DIELECTRICS ELECTRIC ENERGY STORAGE ELECTRIC CURRENTS AND RESISTANCE DC CIRCUITS MAGNETISM SOURCES OF MAGNETIC FIELD ELECTROMAGNETIC INDUCTION AND FARADAY S LAW INDUCTANCE ELECTROMAGNETIC OSCILLATIONS AND AC CIRCUITS MAXWELL S EQUATIONS AND ELECTROMAGNETIC WAVES LIGHT REFLECTION AND REFRACTION LENSES AND OPTICAL INSTRUMENTS THE WAVE NATURE OF LIGHT INTERFERENCE DIFFRACTION AND POLARIZATION SPECIAL THEORY OF RELATIVITY EARLY QUANTUM THEORY AND MODELS OF THE ATOM QUANTUM MECHANICS QUANTUM MECHANICS OF ATOMS MOLECULES AND SOLIDS NUCLEAR PHYSICS AND RADIOACTIVITY NUCLEAR ENERGY EFFECTS AND USES OF RADIATION ELEMENTARY PARTICLES ASTROPHYSICS AND COSMOLOGY Market Description This book is written for readers interested in learning the basics of physics **Physics** Art Hobson,1999 Emphasizes modern physics in a philosophical cultural as well as scientific context Atoms and the structure of matter Speed velocity and acceleration The connections between force mass and acceleration Energy efficiency and electric power The second law of thermodynamics Entropy The automobile and the steam electric generating plant General relativity and cosmology the large scale geometry density and fate of the universe along with the inflationary theory s predictions and the search for the possibly missing mass in the universe The search for extraterrestrial intelligence Quantum theory the electron double slit experiment and the evidence for Bell s interconnectedness principle **Errorless 37 Years NTA NEET**

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