

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

## CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

### Covalent Bonding

#### Section 9.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	$\pi$ bond
---------------	----------	------------	------------	------------

When sharing of electrons occurs, the attachment between atoms that results is called a(n) **(1) covalent bond**. When such an attachment is formed, bond dissociation energy is released, and the process is **(2) exothermic**. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) **(3) molecule**. If the electrons shared are contained between the two atoms, the attachment is called a(n) **(4) sigma bond**. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) **(5)  $\pi$  bond**.

In your textbook, read about single and multiple bonds and bond strengths.

Circle the letter of the choice that best completes the statement or answers the question.

- Is what form do elements such as hydrogen, nitrogen, and oxygen normally occur?
  - a single atom
  - a molecule containing three atoms**
  - a molecule containing two atoms
  - a molecule containing four atoms
- How many electrons are shared in a double covalent bond?
  - none
  - one
  - two
  - four**
- Bond length is the distance between
  - two molecules of the same substance
  - the nuclei of two attached atoms**
  - the electrons in two attached atoms
  - the orbitals of two attached atoms
- Which of the following relationships relating to bond length is generally correct?
  - As the character of the bond, the stronger the bond.**
  - As the character of the bond, the weaker the bond.
  - As the character of the bond, the fewer the electrons in it.
  - As the character of the bond, the lower the bond dissociation energy.

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Chemistry: Matter and Change • Chapter 9

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## CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

#### Section 9.2 Naming Molecules

In your textbook, read about how binary compounds and acids are named from their formulas.

For each statement below, write *true* or *false*.

- Binary molecular compounds are generally composed of a metal and a nonmetal.  
**false**
- The second element in the formula of a binary compound is named using the suffix *-ite*.  
**false**
- The prefix *mono-* indicates three atoms.  
**false**
- The prefix *kilo-* indicates six atoms.  
**true**
- In naming the first element in a formula, the prefix *mono-* is not used.  
**true**
- For binary acids, the hydrogen part of the compound is named using the prefix *hydro-*.  
**true**
- An oxyacid contains only two elements.  
**false**
- If the name of the cation of an oxyacid ends in *-ite*, the acid name contains the suffix *-ate*.  
**false**

In your textbook, read about naming molecular compounds and oxyacids.

For each item in Column A, write the letter of the matching item in Column B.

- | Column A                   | Column B             |
|----------------------------|----------------------|
| c. $\text{CO}$             | a. hydrobromic acid  |
| f. $\text{CO}$             | b. nitrogen trioxide |
| g. $\text{H}_2\text{CO}_3$ | c. carbon monoxide   |
| h. $\text{NO}_2$           | d. nitrous acid      |
| b. $\text{N}_2\text{O}_5$  | e. acetone           |
| d. $\text{HNO}_3$          | f. nitric acid       |
| i. $\text{HNO}_3$          | g. carbonic acid     |
| a. $\text{HBr}$            | h. bromic acid       |
| h. $\text{HBrO}_3$         | i. carbon dioxide    |

Study Guide for Content Mastery

Chemistry: Matter and Change • Chapter 9

# Chemistry Study Guide Content Mastery Teacher

**California High School Teachers'  
Association**



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*Chemistry* ,2001-04-01      Chemistry: Matter & Change, Study Guide For Content Mastery, Student Edition McGraw Hill,2001-03-30 Study Guide and Reinforcement Worksheets allow for differentiated instruction through a wide range of question formats There are worksheets and study tools for each section of the text that help teachers track students progress toward understanding concepts Guided Reading Activities help students identify and comprehend the important information in each chapter      **Resources in Education** ,1997      **High School Chemistry Teachers Magazine** ,1973      *Teachers College Record* ,1924      Teachers College Record James Earl Russell,1924      *Dissertation Abstracts International* ,1980      **The High School Teacher** ,1925      *Research in Education* ,1971      **Contribution to Education** George Peabody College for Teachers,1929      **Glencoe Science** McGraw-Hill Staff,2001-08      **A Study of a Systematic Method of Teaching** Benjamin Foster Mitchell,1929      Proceedings of Annual Meeting California High School Teachers' Association,1919      **School Science and Mathematics** ,1921      **Manual and Course of Study for the High Schools of Alaska** ,1929      The Teaching of Science in Normal Schools and Teachers Colleges Alice Maria Van de Voort,1928      **Comprehensive Dissertation Index, 1861-1972: Education** Xerox University Microfilms,1973      *Report* International Clearinghouse on Science and Mathematics Curricular Developments,1968      *The Science Teachers Bulletin* ,1936      **A Study of Engineering Education** Charles Riborg Mann,1918

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15 basic designs 11 of which were for the martin house itself

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web art glass or light screens as wright called them are a prominent feature of the home the martin house estate contains  
sixteen individual abstract patterns of art glass more than for any other commission during this period the most iconic of  
these is the design popularly known as the tree of life

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laylights and skylights designed by frank lloyd wright exclusively for the martins and their multi residential estate the panels  
were removed when the estate laid vacant and in disrepair from 1937 1954

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web january 26 2015 frank lloyd wright was arguably one of america s premier designers of art glass at the darwin d martin  
house complex alone wright executed nearly 400 examples of decorative glass windows doors skylights laylights sidelights  
and light fixtures in sixteen primary patterns all of which were fabricated by the linden

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web the 394 pieces of art glass contain eleven interrelated primary art glass themes some of which are artificially lit from behind the art glass combines detailed highly stylized organic designs in colored glass to maintain privacy coupled with large panels of clear glass to bring in the surrounding outdoors

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web of particular significance are the nearly 400 art glass windows throughout the complex that act as light screens to visually connect exterior views with the spaces within more patterns of art glass were designed for the martin house than

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web seven art glass windows that replicate architect frank lloyd wright s geometric patterned stain glass have joined two others in the darwin martin house s reception room tree of life art glass

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web this distinctive structure which was commissioned by a ds hollyhock house art glass 137 00 aline barnsdall commissioned wright to build one of his greatest architectural achievements atop a hill in hollywood the hollyhock house in honor of her favorite flower the geometric pattern found ds thomas entry light art glass 124 00

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web martin house maintains preserves and exhibits a fine collection of works of art designed or selected by frank lloyd wright for the martins as well as objects related to the family and the history of their home a historic furnishings plan was completed in 2008

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