

IONIC AND METALLIC BONDING

Chapter Test A

A. Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

Column A

E 1. electron dot structure
E 2. ionic compound
H 3. valence electron
C 4. ionic bond
J 5. chemical formula
G 6. halide ion
A 7. alloy
I 8. octet rule
K 9. formula unit
B 10. coordination number
D 11. metallic bond

Column E

- a. a mixture of two or more elements, at least one of which is a metal
- b. the number of ions of opposite charge surrounding each ion in a crystal
- c. the force of attraction binding oppositely charged ions together
- d. the attraction of valence electrons for positive metal ions
- e. a depiction of valence electrons around the symbol of an element
- f. compound of cations and anions
- g. an anion of a halogen
- h. an electron in the highest occupied energy level of an atom
- i. Atoms in most compounds tend to achieve the electron configuration of a noble gas.
- j. shows the kinds and numbers of atoms in the smallest representative unit of a substance
- k. lowest whole-number ratio of ions in an ionic compound

B. Multiple Choice

Choose the best answer and write its letter on the line.

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C Cleary

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Cracking the SAT Chemistry Subject Test, 2013-2014 Edition Princeton Review, Theodore Silver, M.D., 2013-03-05 Offers test strategies reviews key concepts of chemistry and provides three full length practice tests with answers and explanations

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