



## PARTICLE THEORY AND STATES OF MATTER

What is the basic structure of matter? Understanding the nature of matter and the arrangement of particles is fundamental. The arrangement and movement of particles explain various properties of the matter.

**Particle Theory** is the idea that all matter is made up of tiny particles. These particles have three important characteristics:

- 1. Particles are very small.** They are too small to be seen with the naked eye. They can only be seen using a microscope.
- 2. Particles are constantly moving.** They are always in motion. The speed of motion depends on the temperature. Particles move faster at higher temperatures.
- 3. Particles attract each other.** There are forces of attraction between particles. These forces are stronger in solids and weaker in gases.

**States of Matter** are the different forms that matter can take. They are determined by the arrangement and movement of particles.

**Solid:** Particles are packed closely together in a regular arrangement. They have strong forces of attraction and vibrate in fixed positions.

**Liquid:** Particles are packed closely together but can move past each other. They have moderate forces of attraction.

**Gas:** Particles are far apart and move freely. They have weak forces of attraction.

**Changes of State** occur when the arrangement and movement of particles change. These changes are called phase transitions.

**Melting:** Solid to liquid. Particles gain energy and move more freely.

**Freezing:** Liquid to solid. Particles lose energy and move more closely together.

**Boiling:** Liquid to gas. Particles gain a lot of energy and move very freely.

**Condensation:** Gas to liquid. Particles lose energy and move more closely together.



11. How does the arrangement of particles change when a solid melts? What happens to the forces of attraction between particles?

10. Complete the diagram showing the arrangement of particles in each state of matter. Label the particles and the forces of attraction.

9. What is the change of state when a solid melts?

8. Why is the density of matter different in each state? Which state has the highest density?

7. How does the arrangement of particles change when a gas is cooled?

1. What is particle theory?

2. List all the states of matter. Describe each state and its properties.

3. How does the arrangement of particles change when a solid melts? What happens to the forces of attraction?

4. The solid state is the most dense. Explain why. Which state has the least density?

5. What happens when a solid melts? What happens when a liquid freezes?

6. How does the arrangement of particles change when a gas is cooled? What happens to the forces of attraction?

# Chapter 13 States Of Matter Guided Reading Answers

**Y Pai**



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