

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY**Solutions****Section 15.1 What are solutions?**

In your textbook, read about the characteristic of solutions.

Underline the terms below just now to complete the passage.

solvent	liquid	solvent
insoluble	solid	solvent

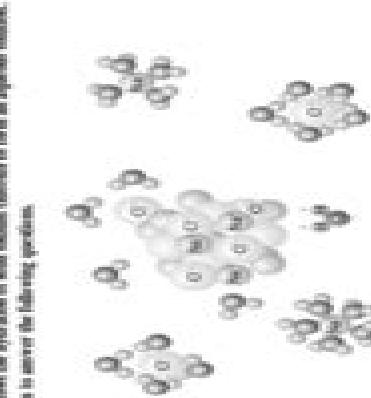
After reading (1) _____ solution _____ of solute is dissolved in solvent.

(2) _____ solute _____ is dissolved in the (3) _____ solvent _____.

(4) _____ solute _____ is dissolved in a solvent, or (5) _____ is in (6) _____ solution. A substance that does not dissolve is (7) _____ insoluble _____ (8) _____ liquid _____ solution or the most common type of solution. If one liquid is soluble in another liquid, such as water and oil, the two liquids are (9) _____ miscible _____ because if one liquid is insoluble in another, the liquids are (10) _____ immiscible _____.

Read about solubility in aqueous solution in your textbook.

Underline the questions to answer the following questions.



1. Indicate in solute is when the solute is in a (1) _____.

(2) _____ is the process of surrounding solute particles with solvent particles to form a solution.

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Section 15.1 content**STUDY GUIDE FOR CONTENT MASTERY****Section 15.1 content**

16. An sodium chloride dissolved in water. What happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

17. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atom). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

18. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solution process to occur.

19. In a glass cup, add the rest of solute may be dissolved, stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent.

In your textbook, read about heat of solution, solubility and factors that affect solubility for each solute and solvent, with type of solute.

True _____ 16. The overall energy change that occurs when a solute dissolves in a solvent is called the heat of solution.

False _____ 17. Solubility is a property of the solute, amount of either the solute or a given amount of solute at a specified temperature and pressure.

False _____ 18. Solubility continues to change as the solute dissolves in a solvent because the heat of solution continues to change.

True _____ 19. In a saturated solution, solute and crystallization are in equilibrium.

True _____ 20. Additional solute can be dissolved in an unsaturated solution, if the solution becomes saturated.

True _____ 21. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Chapter 4 Study Guide For Content Mastery

**Judith L. Irvin, Douglas R.
Buehl, Barbara J. Radcliffe**

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Excel Science Study Guide, Years 9-10 Will Marchment,2004

The book contains coverage of five major topic areas in the NSW School Certificate test: Energy, Force and Motion, Atoms, Elements and Compounds, Structure and Function of Living Things, Earth and Space, Ecosystems, Resources and Technology. A chapter on Investigations and Problem Solving in Science to help with practical skills revision, questions and chapter tests to help you remember important information. A glossary and summary in each section of the book, diagrams and illustrations to help your understanding. A section to help you prepare for the School Certificate test, a sample School Certificate test paper with answers, answers to all questions.

Study Guide to Accompany Introduction to Critical Care Nursing Dawn Ellen Zimmerman, Mary Lou Sole, Marilyn L. Lamborn, Jeanette Hartshorn, 2000-11

Concept maps for each chapter in the parent text help your students sharpen their critical thinking skills and numerous multiple choice review questions challenge their knowledge of each chapter's content. Additional exercises such as crossword puzzles also facilitate learning and key point summaries bring it all together.

Instructor's Resource Manual for Gaines and Miller's Criminal Justice in Action, the Core, Third Edition Gerald P. Fisher,2006

A Study in the Adaptation of the Chicago Unit Plan to Certain Phases of Social Science Work in the Primary Grades Ida B. Helphinstine,1930

Resources in Education ,1996-04

Glencoe Science McGraw-Hill Staff,2001-09

Science Voyages Glencoe,1999-05

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Durand,Vincent Mark Durand,David H. Barlow,2004-08 *Strategies to Enhance Literacy and Learning in Middle School Content Area Classrooms* Judith L. Irvin,Douglas R. Buehl,Barbara J. Radcliffe,2007 This is the only book on the market that focuses specifically on content area reading for the middle grades The third edition of this unique resource has been thoroughly updated to include the most current research in the field of Middle School Literacy Unlike most texts that ignore the middle school reader this book addresses the issues that affect middle school students and teachers and their experiences with literacy instruction Readable and teacher friendly Reading and the Middle School Student provides not only a strong research base but also practical teaching strategies for teachers in all of the content areas This book is designed to be a companion book to Rycik and Irvin *Teaching Reading in the Middle Grades* which focuses on reading in English Language arts classes This book focuses on content area reading instruction Take a Glimpse Inside the Third Edition A wealth of current student examples of strategies for middle grade students for instant use in the classroom New issues and trends facing adolescent literacy including policy and position statements and federal action New ELL emphasis in every chapter outlining specific strategies that can be used by middle school teachers with their English language learners Unique focus on classroom implementation of literacy integrated with content area instruction About Your Authors Judith L Irvin is currently a Professor at Florida State University and serves as the Executive Director of the National Literacy Project She has written and edited numerous books chapters and articles on adolescent literacy Douglas R Buehl is a reading specialist at Madison East High School and District Adolescent Literacy Support Teacher Madison Wisconsin He is Past President of the IRA Secondary Reading Interest Group and has published numerous articles on adolescent literacy Barbara J Radcliffe is an eighth grade reading language arts teacher at Fairview Middle School in Tallahassee Florida Barbara also teaches Teaching English in the Middle School and *Teaching Reading in Secondary English* at Florida State University **Teaching Study Habits and Skills** Ralph Clausius Preston,1959 Instr Resrce Manl-Juvenile Delinquency Siegel,Welsh,Senna,2005-03

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Kiss,2017-02-28 A gold mine of practical easy to use teaching methods strategies and tips to improve learning outcomes for students who score below proficiency levels This fully revised and updated third edition of Teaching Kids with Learning Difficulties in Today's Classroom provides information on integrated learning problem solving and critical thinking in line with Common Core State Standards and 21st century skills It reflects the use of technology and schoolwide cluster grouping in support of all students and includes proven practical classroom tested strategies and step by step instructions for how to use them Sidebars throughout highlight special information for working with students on the autism spectrum tech tips describe technologies that are especially useful for kids with LD Digital content includes all of the book's customizable forms additional content organization charts and a PDF presentation for book study groups and professional development

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