

CHAPTER 9 REVIEW**Stoichiometry****SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. _____ The coefficients in a chemical equation represent the
 - (a) masses in grams of all reactants and products.
 - (b) relative number of moles of reactants and products.
 - (c) number of atoms of each element in each compound in a reaction.
 - (d) number of valence electrons involved in a reaction.
2. _____ Which of the following would not be studied within the topic of stoichiometry?
 - (a) the mole ratio of Al to Cl in the compound aluminum chloride
 - (b) the mass of carbon produced when a known mass of sucrose decomposes
 - (c) the number of moles of hydrogen that will react with a known quantity of oxygen
 - (d) the amount of energy required to break the ionic bonds in CaF_2
3. _____ A balanced chemical equation allows you to determine the
 - (a) mole ratio of any two substances in the reaction.
 - (b) energy released in the reaction.
 - (c) electron configuration of all elements in the reaction.
 - (d) reaction mechanism involved in the reaction.
4. _____ The relative number of moles of hydrogen to moles of oxygen that react to form water represents a(n)
 - (a) reaction sequence.
 - (b) bond energy.
 - (c) mole ratio.
 - (d) element proportion.

5. Given the reaction represented by the following unbalanced equation:

$$\text{N}_2\text{O}(g) + \text{O}_2(g) \rightarrow \text{NO}_2(g)$$

- a. Balance the equation.

- b. What is the mole ratio of NO_2 to O_2 ?
- c. If 20.0 mol of NO_2 form, how many moles of O_2 must have been consumed?
- d. Twice as many moles of NO_2 form as moles of N_2O are consumed. True or False?
- e. Twice as many grams of NO_2 form as grams of N_2O are consumed. True or False?

Chapter 9 Reading Guide Stoichiometry

T. R. Dickson

Chapter 9 Reading Guide Stoichiometry:

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills,Amina El-Ashmawy,2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Study Guide to Accompany Basics for Chemistry Martha Mackin,2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter

contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the

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equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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Kimble,1994

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Preparation Services,2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter s end Two full length practice tests with detailed answer explanations are included in the book

Study Guide for Chemistry by Steven S. Zumdahl Martha B. Barrett,1986 **Student Study Guide to Accompany Petrucci's General Chemistry, 3rd. Ed**

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Wismer,1985 *Introduction to Chemistry, Study Guide* T. R. Dickson,1999-07-07 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and chemical vision and refining problem solving skills Many chapters in this edition have been rewritten and rearranged to vitalize the topics and to include interesting examples analogies and images Study Guide General Chemistry, Second Edition, Becker/Wentworth Becker,Benjamin L. Carroll,1980 **Merrill**

Chemistry Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 Student Study Guide for Introduction to Chemistry John Dickson,1979-01-25

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- Chapter 2: Essential Elements of Chapter 9 Reading Guide Stoichiometry
- Chapter 3: Chapter 9 Reading Guide Stoichiometry in Everyday Life
- Chapter 4: Chapter 9 Reading Guide Stoichiometry in Specific Contexts
- Chapter 5: Conclusion

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