

CHEM 101 Final Exam with complete solutions

How many significant figures are in this number, 0.01160 - ANSWER: → 4

Select the correct scientific notation for 117000 - ANSWER: → 1.17×10^5

the temperature of an experiment is designed at 122 degrees F. What is it in Celsius? - ANSWER: → 50 degrees Celsius

An experiment is performed at -173 degrees C. What is the temperature in Kelvin? - ANSWER: → 100K

A piece of metal has a mass of 340 g and a volume of 17.6 mL. What is the density of the metal? - ANSWER: → 19.3 g/mL

The comparison of the density of a substance with the density of water is defined as _____. - ANSWER: → specific gravity

The following forms of energy can be classified as potential energy EXCEPT _____.

- A). Gasoline
- B). Running
- C). Peanut butter sandwich
- D). Water held by a dam - ANSWER: → running

The following examples are chemical changes EXCEPT

- A). Rusting iron
- B). Cooking an egg
- C). Burning gasoline
- D). Melting ice - ANSWER: → melting ice

The amount of heat necessary to raise the temperature of 1 g liquid water by 1oC, is

defined as _____. - ANSWER: → a calorie

Chem 101 Final Exam Review

IM Harris

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