

10**CHEMICAL QUANTITIES****Practice Problems**

On a piece of paper, solve the following problems.

YOU MUST SHOW YOUR WORK TO GET CREDIT FOR YOUR ANSWERS!!!

SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER

1. What is the molar mass of sucrose ($C_{12}H_{22}O_{11}$)?
2. What is the molar mass of each of the following compounds?
 - a. phosphorus pentachloride (PCl_5)
 - b. uranium hexafluoride (UF_6)
3. Calculate the molar mass of each of the following ionic compounds:
 - a. $KMnO_4$
 - b. $Ca_3(PO_4)_2$
4. How many moles is 3.52×10^{24} molecules of water?
5. How many atoms of zinc are in 0.60 mol of zinc?
6. What is the mass of 1.00 mol of oxygen (O_2)?

SECTION 10.2 MOLE–MASS AND MOLE–VOLUME RELATIONSHIPS

1. What is the molar mass of each of the following compounds?
 - a. $C_6H_{12}O_6$
 - b. $NaHCO_3$
 - c. C_3H_{12}
 - d. KNH_4SO_4
2. Calculate the mass in grams of each of the following:
 - a. 8.0 mol lead oxide (PbO)
 - b. 0.75 mol hydrogen sulfide (H_2S)
 - c. 0.00100 mol silicon tetrahydride (SiH_4)
 - d. 1.50×10^{-2} mol molecular oxygen (O_2)
 - e. 2.30 mol ethylene glycol ($C_2H_6O_2$)
3. How many grams are in 1.73 mol of dinitrogen pentoxide (N_2O_5)?
4. How many grams are in 0.658 mol of calcium phosphate [$Ca_3(PO_4)_2$]?
5. Calculate the number of moles in each of the following:
 - a. 0.50 g sodium bromide ($NaBr$)
 - b. 13.5 g magnesium nitrate [$Mg(NO_3)_2$]
 - c. 1.02 g magnesium chloride ($MgCl_2$)
 - d. 0.00100 g monochloromethane (CH_3Cl)
 - e. 1.50×10^{-3} g propylene glycol [$C_3H_8(OH)_2$]
6. A chemist plans to use 435.0 grams of ammonium nitrate (NH_4NO_3) in a reaction. How many moles of the compound is this?
7. A solution is to be prepared in a laboratory. The solution requires 0.0465 mol of quinine ($C_{20}H_{24}N_2O_2$). What mass, in grams, should the laboratory technician obtain in order to make the solution?
8. What is the volume at STP of 2.66 mol of methane (CH_4) gas?
9. How many moles is 135 L of ammonia (NH_3) gas at STP?

Chemical Quantities Guide Answers

AW Rasmussen



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Programmed Learning and Individually Paced Instruction Carl H. Hendershot,1973 **Physical and Chemical Hydrogeology** Patrick A. Domenico,Franklin W. Schwartz,1997-10-15 This book shows readers how to apply hydrogeology principles to a host of problems related to water supply contamination and energy resources It discusses hydraulic testing modeling of contaminant transport process and parameter determination and remediation It also addresses porosity permeability and flow for continental environments marine environments and the borders between them *Chemical Age* ,1922 **Chemical Analysis of the Environment and Other Modern Techniques** Sut Ahuja,2012-12-06 With the rise in general awareness of the effects of trace chemicals in the environment on man s health it has been realized that traditional methods of analysis are often inadequate Reliable analyses are needed in the fractional parts per million range of contaminants in condensed phases and of the order of micrograms per cubic meter in air Trying to get meaningful answers regarding such minute amounts raises cogent problems in all stages of an analysis It is most appropriate therefore that the 1971 Eastern Analytical Symposium should have four half day sessions devoted to this general field Two of these entitled Trace Metals in the Envi ronment were assembled by Dr Kneip one on Pesticides in the Environment Recently Discovered Analytical Problems by Dr Zweig and one on The Determination of Anions in Water by Dr Lambert Together these reports furnish a fairly complete picture of the present state of environmental anal ysis The remainder of this volume is devoted to pharmaceutical analysis a diversified field in which nearly all analytical methods find a place Partly because of this multiplicity of techniques and partly due to the large number of samples which must be examined in connection with the manu facture biological testing and clinical applica tion of pharmaceutical preparations this area is particularly appropriate for the introduction of automation The objective broadly is to speed up multiple analyses without the sacrifice of accuracy

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