

**Balancing and Reactions Test Unit 2 SCH3U****Name:**

    K       T       C       A   **Assessment of Learning (Product)**  
**15      20      5      4**

Multiple Choice: Circle the most correct answer. (10K)

1. Propane gas,  $C_3H_8$ , burns in excess oxygen gas. When the equation for this reaction is correctly balanced and all coefficients are reduced to their lowest whole-number terms, the coefficient for  $H_2O$  is...

- A) 1    B) 2    C) 3    D) 4    E) 5    f) 6

2. State the type of reaction:  $2 Na + 2 H_2O \rightarrow 2 NaOH + H_2$

- A) synthesis                      B) decomposition              C) single displacement  
D) double displacement        E) combustion                  F) no reaction

3. Which of the following types of chemical reactions produces water and an ionic compound?

- A) single displacement              B) neutralization              C) decomposition              D) synthesis

4. State the type of reaction:  $Ca(NO_3)_2 + 2 HCl \rightarrow CaCl_2 + 2 HNO_3$

- A) synthesis                      B) decomposition              C) single displacement  
D) double displacement        E) combustion                  F) no reaction

5. If chlorine gas is produced by single displacement, the other halogen in the reaction must be

- A) bromine    B) iodine    C) astatine    D) fluorine

6. In what kind of reaction do two or more substances combine to form a new compound?

- A) decomposition reaction              C) double-displacement reaction  
B) single displacement                  D) synthesis reaction

7. A group whose atoms (M) undergo the reaction:



- A) group 1              B) group 2              C) group 13              D) group 14              E) group 15

8. When heated, a metal carbonate decomposes into a metal oxide and

- A) carbon.              B) carbon dioxide              C) oxygen.              D) hydrogen.

9. Oxides of active metals, such as  $CaO$ , react with water to produce

- A) metal carbonates.              B) metal hydrides.              C) acids.              D) metal hydroxides.

10. Which of the following statements is true about single-displacement reactions?

- A) They are restricted to metals.              C) Two reactants produce two products.  
B) They involve a single product.              D) Any metal replaces any other metal.

# Chemical Reactions Test Answers

**Mark Steyvers**



## **Chemical Reactions Test Answers:**

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**Partial Solutions Manual** Darrell D. Ebbing, David Bookin, 1998

**CliffsNotes AP Chemistry** Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book

*How to Pass Dental Aptitude Tests, Questions and Answers* College Publishing Corporation, Brooklyn, 1962

Instructional Tests in Chemistry Earl Rouse Glenn, Louis E. Welton, 1929

McGraw-Hill's SAT Subject Test Chemistry, 3rd Edition Thomas Evangelist, 2011-02-03

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**The Photographic News**, 1886

*Johnson's New Universal Cyclopædia* Frederick Augustus Porter Barnard, Arnold Guyot, 1875

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**Table of Contents Chemical Reactions Test Answers**

1. Understanding the eBook Chemical Reactions Test Answers
  - The Rise of Digital Reading Chemical Reactions Test Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Reactions Test Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemical Reactions Test Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Reactions Test Answers
  - Personalized Recommendations
  - Chemical Reactions Test Answers User Reviews and Ratings
  - Chemical Reactions Test Answers and Bestseller Lists
5. Accessing Chemical Reactions Test Answers Free and Paid eBooks
  - Chemical Reactions Test Answers Public Domain eBooks
  - Chemical Reactions Test Answers eBook Subscription Services
  - Chemical Reactions Test Answers Budget-Friendly Options
6. Navigating Chemical Reactions Test Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemical Reactions Test Answers Compatibility with Devices
  - Chemical Reactions Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemical Reactions Test Answers
  - Highlighting and Note-Taking Chemical Reactions Test Answers
  - Interactive Elements Chemical Reactions Test Answers
8. Staying Engaged with Chemical Reactions Test Answers



- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemical Reactions Test Answers
9. Balancing eBooks and Physical Books Chemical Reactions Test Answers
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemical Reactions Test Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemical Reactions Test Answers
- Setting Reading Goals Chemical Reactions Test Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemical Reactions Test Answers
- Fact-Checking eBook Content of Chemical Reactions Test Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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