

Atomic Number

- \_\_\_\_\_ across a period
- \_\_\_\_\_ down a family

Atomic Mass

- \_\_\_\_\_ across a period
- \_\_\_\_\_ down a family

Atomic Radius

\_\_\_\_\_ the distance between \_\_\_\_\_ of two like \_\_\_\_\_.

- \_\_\_\_\_ across a period (due to increased attraction of more positive nuclei to electron clouds)
- \_\_\_\_\_ down a family (due to addition of an energy level)

Electron Affinity

amount of \_\_\_\_\_ when an \_\_\_\_\_ gains an \_\_\_\_\_

- \_\_\_\_\_ across a period
- \_\_\_\_\_ down a family

Ionization Energy

amount of \_\_\_\_\_ required to \_\_\_\_\_ an \_\_\_\_\_ from a neutral \_\_\_\_\_

- \_\_\_\_\_ across a period
- \_\_\_\_\_ down a family

Electron Distribution - Forming Ions

- Atoms become ions by \_\_\_\_\_ or \_\_\_\_\_ electrons to become \_\_\_\_\_.
- An atom is stable when it has a \_\_\_\_\_ outer energy level (\_\_\_\_\_ valence electrons).
- Atoms that lose (give away) electrons become \_\_\_\_\_ charged ions. Atoms that gain (steal) electrons become \_\_\_\_\_ charged ions.

**The Chemistry Quiz**

CR1.

CR2.

1.

2.

3.

4.

5.

# Chemistry Episode Note Taking Guide Key

**Georgia Woodfield, Benedict Lyle  
Phillips, Victoria Taylor, Amy  
Hawkins, Andrew Stanton**



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