



CHEMISTRY

CHEAT SHEET

EQUILIBRIUM

EQUILIBRIUM CHEAT SHEET

Chemical Equilibrium occurs when the concentrations of the chemical species remain the same with time (dynamic equilibrium). Equilibrium reactions are represented by the symbol \rightleftharpoons .

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| <p>What is Equilibrium?</p> <p>Chemical Equilibrium occurs when the concentrations of the chemical species remain the same with time (dynamic equilibrium). Equilibrium reactions are represented by the symbol \rightleftharpoons.</p> <p>For a general reaction:</p> $aA + bB \rightleftharpoons cC + dD$ <p>The equilibrium constant is given by:</p> $K_c = \frac{[C]^c [D]^d}{[A]^a [B]^b}$ | <p>Equilibrium Constant</p> <p>For a general reaction:</p> $aA + bB \rightleftharpoons cC + dD$ <p>The equilibrium constant is given by:</p> $K_c = \frac{[C]^c [D]^d}{[A]^a [B]^b}$ |  |
| <p>Equilibrium Constant, K_c</p> <p>For a general reaction:</p> $aA + bB \rightleftharpoons cC + dD$ <p>The equilibrium constant is given by:</p> $K_c = \frac{[C]^c [D]^d}{[A]^a [B]^b}$ | <p>Equilibrium Constant, K_p</p> <p>For a general reaction:</p> $aA + bB \rightleftharpoons cC + dD$ <p>The equilibrium constant is given by:</p> $K_p = \frac{P_C^c P_D^d}{P_A^a P_B^b}$ | |
| <p>Equilibrium Constant, K_p</p> <p>For a general reaction:</p> $aA + bB \rightleftharpoons cC + dD$ <p>The equilibrium constant is given by:</p> $K_p = \frac{P_C^c P_D^d}{P_A^a P_B^b}$ | <p>Le Chatelier's Principle: the effect of changing conditions on equilibrium</p> | |
| <p>Le Chatelier's Principle</p> <p>When a system at equilibrium is subjected to a change in concentration, pressure, or temperature, the system will adjust itself to counteract the change and restore a new equilibrium.</p> | <p>Le Chatelier's Principle</p> <p>When a system at equilibrium is subjected to a change in concentration, pressure, or temperature, the system will adjust itself to counteract the change and restore a new equilibrium.</p> | |

Chemistry Equilibrium Study Guide Answers

G. Marc Loudon, Joseph G. Stowell



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