

CHEMISTRY CHEAT SHEET

NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS Cheat Sheet

Naming Compounds with Two Monatomic Ions

1. **Identify the cation and anion.**
2. **Find the cation's name.**
3. **Find the anion's name.**
4. **Drop the suffix -ide from the anion's name.**
5. **Drop the prefix poly- from the anion's name.**
6. **Drop the prefix hetero- from the anion's name.**
7. **Drop the prefix homo- from the anion's name.**
8. **Drop the prefix homo- from the anion's name.**
9. **Drop the prefix homo- from the anion's name.**
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Common Polyatomic Ions and their Names

| Common Polyatomic Ions and their Names | |
|--|-----------|
| OH ⁻ | Hydroxide |
| Cl ⁻ | Chloride |
| Br ⁻ | Bromide |
| I ⁻ | Iodide |
| SO ₄ ²⁻ | Sulfate |
| CO ₃ ²⁻ | Carbonate |
| NO ₃ ⁻ | Nitrate |
| PO ₄ ³⁻ | Phosphate |
| SO ₃ ²⁻ | Sulfite |
| NO ₂ ⁻ | Nitrite |
| ClO ₄ ⁻ | Chlorate |
| BrO ₃ ⁻ | Bromate |
| IO ₃ ⁻ | Iodate |
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| BrO ₃ ⁻ | Bromate |
| IO ₃ ⁻ | Iodate |

Common Monatomic Ions and their Names

| Common Monatomic Ions and their Names | |
|---------------------------------------|------------|
| H ⁺ | Hydron |
| Li ⁺ | Lithium |
| Na ⁺ | Sodium |
| K ⁺ | Sodium |
| Ca ²⁺ | Calcium |
| Mg ²⁺ | Magnesium |
| Al ³⁺ | Aluminum |
| Si ⁴⁺ | Silicon |
| P ⁵⁺ | Phosphorus |
| S ⁶⁺ | Sulfur |
| Cl ⁻ | Chloride |
| Br ⁻ | Bromide |
| I ⁻ | Iodide |
| OH ⁻ | Hydroxide |
| SO ₄ ²⁻ | Sulfate |
| CO ₃ ²⁻ | Carbonate |
| NO ₃ ⁻ | Nitrate |
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| IO ₃ ⁻ | Iodate |

Chemistry Naming Compounds Study Guide

Darrell D. Ebbing

Chemistry Naming Compounds Study Guide:

Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein,2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems

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Study Guide for Chemical Principles [by] Steven S. Zumdahl Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples

Study Guide, Introductory Chemistry, a Foundation ; Introductory Chemistry ; Basic Chemistry Iris Stovall, Donald J. DeCoste, 2000 Chemistry, Student Study Guide John A. Olmsted, Gregory M.

Williams, 2005-02-02 100% Pure Chemical Understanding Every morning many of us are energized by a cup of coffee Imagine if you were as energized by understanding the chemistry in your morning cup from the coffee trees which fill red coffee berries with caffeine and a variety of other chemical substances to the feathery crystals formed by the caffeine molecules to the decaffeinating machines which use liquid solvents to remove this stimulant from some of the beans Now that's real chemical understanding Olmsted and Williams Fourth Edition of Chemistry focuses on helping you see and think about the world and even your coffee as a chemist This text helps you understand how chemical phenomena are governed by what happens at the molecular level apply critical thinking skills to chemical concepts and problems and master the basic mathematical techniques needed for quantitative reasoning You'll see the world as chemists do and learn to appreciate the chemical processes all around us A Fourth Edition with a lot of new perks Revisions include a new early energy chapter revised coverage of bonding expanded coverage of intermolecular forces and increased coverage of multiple equilibria including polyprotic acids New pedagogy strengthens students critical thinking and problem solving skills Visual Summaries at the end of each chapter use molecular and diagrammatic visual elements to summarize essential skills concepts equations and terms eGrade Plus provides an integrated suite of teaching and learning resources including a complete online version of the text links between problems and relevant sections in the online text practice quizzes the Visual Tutor Interactive LearningWare problems and lab demos as well as homework management and presentation features for instructors

Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section

entitled Tying It All Together with a Laboratory Application

Organic Chemistry Study Guide and Solutions Marc Loudon, Jim Parise, 2015-07-01 Parise and Loudon's Study Guide and Solutions Manual offers the following learning aids:

Links that provide hints for study approaches to problem solving and additional explanations of challenging topics.

Further Explorations that provide additional depth on key topics.

Reaction summaries that delve into key mechanisms and stereochemistry.

Solutions to all the textbook problems. Rather than providing just the answer, many of the solutions provide detailed explanations of how the problem should be approached.

Solutions Manual and Study Guide to Accompany Introduction to Organic Chemistry, 4th Ed Paul A. Bartlett, Judith G. Koch, 1992

Study Guide for General Chemistry and College Chemistry, Eighth Editions by Holtzclaw and Robinson Norman E. Griswold, Henry Fuller Holtzclaw, 1988

General Chemistry, Study Guide James E. Brady, 1990-03-01 The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated, reorganized, and streamlined. Changes include more accessible introductory chapters with greater stress on the logic of the periodic table, earlier introduction of redox reactions, greater emphasis on the concept of energy, a new section on Lewis structures, earlier introduction of the ideal gas law, and a new development of thermodynamics. Each chapter ends with review questions and problems.

General Chemistry Study Guide Sixth Edition Darrell D. Ebbing, 1999

Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter.

Study Guide for Introductory Chemistry Iris Stovall, 1990

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