



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the ions involved in the compound)



Directions for naming:

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.
3. If the cation is a transition metal, include the oxidation state in Roman numerals in parentheses.
4. If the anion is a polyatomic ion, use its name.
5. If the compound is an acid, use the prefix H- and the name of the anion.
6. If the compound is a base, use the name of the cation and hydroxide.

Binary Compounds

First Element	Anion
H	hydrogen
C	carbon
N	nitrogen
O	oxygen
F	fluorine
Cl	chlorine
S	sulfur
Br	bromine
I	iodine
Al	aluminum
Zn	zinc
Fe	iron
Cu	copper
Pb	lead
Ag	silver

Common Ions and Their Names

Formula	Name
Na^+	Sodium
K^+	Potassium
Ca^{2+}	Calcium
Mg^{2+}	Magnesium
Al^{3+}	Aluminum
Zn^{2+}	Zinc
Fe^{2+}	Iron (II)
Fe^{3+}	Iron (III)
Cu^+	Copper (I)
Cu^{2+}	Copper (II)
Pb^{2+}	Lead (II)
Pb^{4+}	Lead (IV)
Ag^+	Silver

Naming Compounds with a Central and Peripheral



Sodium Chloride

Directions for naming:

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.
3. If the cation is a transition metal, include the oxidation state in Roman numerals in parentheses.
4. If the anion is a polyatomic ion, use its name.

Naming Acids

Formula	Central Element	Formula	Central Element
HCl	Hydrogen	HNO_3	Nitrogen
HClO_4	Hydrogen	HNO_2	Nitrogen
H_2SO_4	Sulfur	H_2SO_3	Sulfur
H_2CO_3	Carbon	H_2CO_2	Carbon
$\text{H}_2\text{C}_2\text{O}_4$	Carbon	$\text{H}_2\text{C}_2\text{O}_3$	Carbon
$\text{H}_2\text{C}_2\text{O}_2$	Carbon	$\text{H}_2\text{C}_2\text{O}_1$	Carbon
$\text{H}_2\text{C}_2\text{O}_0$	Carbon	$\text{H}_2\text{C}_2\text{O}_{-1}$	Carbon

Common Names for Specific Compounds

Formula	Common Name	Formula	Common Name
NaCl	Table salt	NaOH	Sodium hydroxide
KCl	Saline	KOH	Potassium hydroxide
H_2SO_4	Sulfuric acid	H_2CO_3	Carbonic acid
HCl	Hydrochloric acid	H_2O	Water

Chemistry Naming Compounds Study Guide

Darrell D. Ebbing



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