

# Chemical Reactions

A N D QUATIONS

## Chemical Equation

Reactants left hand side → Products Right hand side

magnesium + Oxygen → magnesium oxide



## BALANCING EQUATIONS

number of atoms of every element before and after reaction is same



### COMBINATION

Two or more substance combine to form a single product.



### DECOMPOSITION

Single reactant breaks up to form two or more products



### DISPLACEMENT

More reactive element removes the other element from the compound and itself forms the compound



### PHYSICAL STATE -



### CHANGE IN HEAT -



### EVOLUTION OF GOS -



### PRECIPITATION -



### DIFF. REACTIONS (CONDITION -



Chemical equations are a way of showing chemical reactions.

LAW OF CONSERVATION OF MATTER  
a chemical equation must be balanced.

## Chemical reactions

Chemical reactions can occurs only:  
↳ when the chemical change.  
↳ Rusting in Iron, Burning of coal.



### DOUBLE DISPLACEMENT

Exchange of ions between the reactants take place.



### REDUCTION

gain of hydrogen or loss of oxygen.



### OXIDATION

gain of oxygen or loss of hydrogen.

oxidation

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# Chemistry Notetaking Guide

**A Gutmann**

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