



# CHEMISTRY

## CHEAT SHEET



# INTERMOLECULAR FORCES

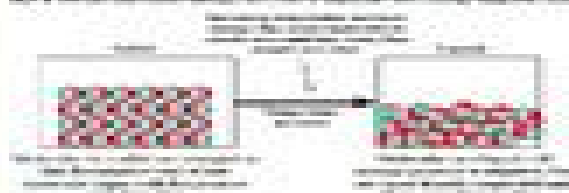
## INTERMOLECULAR FORCES CHEAT SHEET

Intermolecular forces are attractive forces between molecules. They are responsible for many physical properties and chemical behavior of substances.

The strength of intermolecular forces depends on the nature of the molecules. The order of intermolecular forces is as follows: Hydrogen bonding > Dipole-dipole interaction > London dispersion forces.

Intermolecular forces	Description	Types of molecules
Hydrogen bonding	Occurs when hydrogen is bonded to highly electronegative atoms (F, O, N). The resulting partial positive charge on hydrogen interacts with the partial negative charge on the electronegative atom.	Small molecules containing F, O, or N
Dipole-dipole interaction	Occurs between polar molecules. The positive end of one molecule is attracted to the negative end of another molecule.	Polar molecules
London dispersion forces	Occurs in all molecules, but is only significant in non-polar molecules. It is caused by temporary dipoles.	All molecules

Intermolecular forces are responsible for many physical properties of substances, such as boiling point, melting point, and viscosity. The strength of intermolecular forces increases with the size and complexity of the molecules.



The order of intermolecular forces is as follows: Hydrogen bonding > Dipole-dipole interaction > London dispersion forces.

	Formula	Boiling Point	Melting Point
Water	$H_2O$	100°C	0°C
Alcohol	$C_2H_5OH$	78°C	-114°C
Gasoline	$C_8H_{18}$	125°C	-95°C

# Chemistry Physics Forces Study Guide

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**Chemistry, Student Study Guide** John A. Olmsted, Gregory M. Williams, 2005-02-02 100% Pure Chemical Understanding Every morning many of us are energized by a cup of coffee Imagine if you were as energized by understanding the chemistry in your morning cup from the coffee trees which fill red coffee berries with caffeine and a variety of other chemical substances to the feathery crystals formed by the caffeine molecules to the decaffeinating machines which use liquid solvents to remove this stimulant from some of the beans Now that's real chemical understanding Olmsted and Williams Fourth Edition of Chemistry focuses on helping you see and think about the world and even your coffee as a chemist This text helps you understand how chemical phenomena are governed by what happens at the molecular level apply critical thinking skills to chemical concepts and problems and master the basic mathematical techniques needed for quantitative reasoning You'll see the world as chemists do and learn to appreciate the chemical processes all around us A Fourth Edition with a lot of new perks Revisions include a new early energy chapter revised coverage of bonding expanded coverage of intermolecular forces and increased coverage of multiple equilibria including polyprotic acids New pedagogy strengthens students critical thinking and problem solving skills Visual Summaries at the end of each chapter use molecular and diagrammatic visual elements to summarize essential skills concepts equations and terms eGrade Plus provides an integrated suite of teaching and learning resources including a complete online version of the text links between problems and relevant sections in the online text practice quizzes the Visual Tutor Interactive LearningWare problems and lab demos as well as homework management and presentation features for instructors *United States Armed Forces Institute Catalog* United States Armed Forces Institute, 1963 **Resources in Education**, 1984 **Catalog of Copyright Entries. Third Series** Library of Congress. Copyright Office, 1954 Includes Part 1A Number 1 Books January June and Part 1B Number 1 Pamphlets Serials and Contributions to Periodicals January June The Medical Student's Manual of Chemistry Rudolph August Witthaus, 1890 Cleaning and Lubricating Louis E. Noble, Joseph A. Roenigk, 1928 *The Literary Guide and Rationalist Review*, 1925 What and how to read: a guide to recent English Literature, containing a classified list of the best books published in England and America during the past five years, etc G. A. F. VAN RHYN, 1875 *Answers to questions prescribed by medical state boards* Robert Borneman Ludy, 1905 *Wiseman Review*, 1902 Transactions of the Kansas State Board of Agriculture Kansas. State Board of Agriculture, 1874 Vols for 1874 76 include also Transactions of the Kansas Academy of Science The Dublin Review Nicholas Patrick Wiseman, 1902 The Brass World and Platers Guide, 1923 Annual Report of the Pennsylvania Department of Agriculture Pennsylvania. Department of Agriculture, 1902 Report of Pennsylvania Forestry Commission published in 1896 1895 pt 2 *The Chemical News and Journal of Industrial Science* William Crookes, James H. Gardiner, Gerald Druce, H. W. Blood-Ryan, 1909 *An Introductory Guide to EC Competition Law and Practice* Valentine Korah, 1994 **The popular encyclopedia; or, 'Conversations Lexicon': [ed. by**

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