

Ch. 9 Reactions in Aqueous Solutions

Molarity and Solution Stoichiometry

- Molarity (M) is used to describe solution concentration because it is very easy to determine the number of moles of solute
 - Solute is what gets dissolved in the solvent
 - Solvent is water (typically)
 - Solution = solute + solvent
- Molarity = (#mol of solute(n)) / (Volume(V) of solution(L))
 - #mol solute = M x V
 - n = M x V
- Four things to know about molarity
 - M x V = # mol of solute
 - Molarity can be used as a conversion factor in stoichiometry problems.
 - You can always replace "M" with (mol/L)
 - An Important Molarity shortcut:

$$M = \frac{\text{# mol Solute}}{\text{L sol'n}} \times 10^{-3}$$

$$M = \frac{\text{# mol Solute}}{\text{L sol'n}}$$

Solution by dilution:

- M₁V₁ = M₂V₂ or M_(concentrated)V_(concentrated) = M_(dilute)V_(dilute)

problems:

- What is the concentration of a solution of HCl that was prepared by diluting 5.00 mL of 12.0 M HCl to 150.0 mL?

$$\frac{M_1V_1}{V_2} = \frac{M_2V_2}{V_2}$$

$$\frac{M_1V_1}{V_2} = M_2$$

$$\frac{(12.0\text{ M})(5.00\text{ mL})}{(150.0\text{ mL})} = 0.400\text{ M}$$

- How many L of 0.100 M HCl are required to completely react with 13.0 g of MgCO₃ in the following reaction?



$$\# \text{ L HCl} = 13.0 \text{ g MgCO}_3 \times \frac{1 \text{ mol MgCO}_3}{84.32 \text{ g MgCO}_3} \times \frac{2 \text{ mol HCl}}{1 \text{ mol MgCO}_3} \times \frac{1 \text{ L}}{0.100 \text{ mol HCl}} = 3.08 \text{ L HCl}$$

Electrolytes

- Be sure to review:
 - Names, formulas, and charges of all ions.
 - These terms: solution, solvent, solute, aqueous solutions
- Light bulb demonstration.

Chemistry 1411 Lab Answers

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Radical Solutions and eLearning Daniel Burgos,2020-05-22 Educational Technology is the right couple to a radical innovation Thanks to the appropriate technology in the right context with the best fit to the target audience education can be drastically improved meaning a better performance competence achievement match with the user's expectations and with the market needs Serious games Virtual reality Augmented reality Remote labs Online learning Blockchain Mobile learning and many other key technologies allow for a better explanation of so many subjects and even more for a complete student involvement and a full teacher engagement into the educational system Technology gives another angle to the same content provides the user with a personalised experience and pushes the limits of knowledge a little further every time This book presents a number of radical innovations through technology from experienced cases studies to be replicated and inspired by a powerful resource handbook for cutting edge education

Evaluated Kinetic Data for High Temperature Reactions

D. L. Baulch,1972 *Synchrotron Radiation* C. Kunz,2013-11-11 Synchrotron radiation as a spectroscopic research tool has undergone a most interesting and astonishing historical development and has now come to the stage of an exciting boom The machines which produce synchrotron radiation were built and developed exclusively for other purposes in the past namely high energy physics At the same time however they involuntarily became better and better light sources for the spectral range from the visible to the hard x ray region Now we are at the point that the first few storage rings have gone into operation as machines dedicated to synchrotron radiation and several more are in the stage of construction and planning All this was brought about by the successful research performed during the past fifteen years in which several groups all over the world have participated at different accelerator centers mostly symbiotic with high energy physics As it happens with a young and rapidly developing field the number of reviews and monographs is still minute The objective of this book is to fill an apparent gap and to provide a sound basis for those who are interested in synchrotron radiation and its applications

Publications United States. National Bureau of Standards,1977 *Evaluated Kinetic Data for High Temperature Reactions: Homogeneous gas phase reactions of the H₂-N₂-O₂ system* ,1972 **Publications of the National Bureau of Standards** United States. National Bureau of Standards,1976 NBS Special Publication ,1968 EPA Publications Bibliography Quarterly Abstract Bulletin United States. Environmental Protection Agency,1994 *A Comprehensive Treatise on Inorganic and Theoretical Chemistry* Joseph William Mellor,1961 **Supplement to Mellor's Comprehensive Treatise on Inorganic and Theoretical Chemistry** Joseph William Mellor,1961 **Reports on the Progress of Applied Chemistry** Society of Chemical Industry (Great Britain),1962 **Supplement to Mellor's Comprehensive Treatise on Inorganic and Theoretical Chemistry: suppl. 3. K, Rb, Cs, Fr** Joseph William Mellor,1961 **Electrogenerative Chemical Processes** Sergei Yurchak,1969 Publications of the National Bureau of Standards ... Catalog United States. National Bureau of Standards,1966 **EPA Publications Bibliography** ,1994 *Kiln-Drying of Lumber* R.B. Keey,T.A.G.

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