

1. Perform the following operations and round your answers to the correct number of significant digits.
a. $2.354 - 2.354$ b. $26.0 \text{ cm}^2 \div 10.1 \text{ cm}$

2. Convert the following and express your answers in scientific notation.
a. 378 Mm to m

b. 3,000,000 mg to kg

3. Water contains 11.2% hydrogen and 88.8% oxygen. What mass of water contains 50.0 g. of oxygen? IIP Answer: 50-50 g H₂O

4. What is the mass of a rectangular object measuring 0.025 m by 2.5 cm by 5.6 cm and has a density of 3.86 g/cm³? IIP Answer: 200 g

5. State the number of neutrons in each of the following
a. Iodine-127 b. ^{75}Ne

6. Silver occurs naturally as ¹⁰⁷Ag and ¹⁰⁹Ag. Given the mass and abundance of silver-109 (108.91 amu and 48.16%), what is the isotopic mass of silver-107? IIP Answer: 107 amu

7. State which of the following are physical or chemical properties or changes
a. sugar dissolved in water
b. Manganese bubbles in acid
c. Silver is malleable

8. What is the total number of orbitals in the $n=2$ energy level? Indicate the type of orbitals and how many of each type of orbital. Indicate the total number of electrons that can reside in $n=2$.

9. Write the electron configuration for Mn²⁺

10. Refer to the periodic table to answer the following:
a. The name of the alkaline earth metal in the third energy level
b. The name and symbol of the element in group VA that is a gas at room temp and pressure

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