

**11****CHEMICAL REACTIONS****Chapter Test B****A. Matching**

Match each term in Column B with the correct description in Column A. Write the letter of the correct term on the line.

**Column A**

- \_\_\_\_\_ 1. a reaction in which an element or compound reacts with oxygen, often producing energy in the form of heat or light
- \_\_\_\_\_ 2. a reaction in which two or more substances react to form a single substance
- \_\_\_\_\_ 3. an equation that indicates only those particles that actually take part in the reaction
- \_\_\_\_\_ 4. a substance that speeds up a reaction without being used up
- \_\_\_\_\_ 5. ions that are not directly involved in a reaction
- \_\_\_\_\_ 6. a reaction in which a single compound is broken down into two or more products
- \_\_\_\_\_ 7. an equation in which each side has the same number of atoms of each element
- \_\_\_\_\_ 8. a reaction in which atoms of an element replace the atoms of a second element in a compound
- \_\_\_\_\_ 9. a list of metals in order of decreasing reactivity
- \_\_\_\_\_ 10. a reaction that involves an exchange of positive ions between two compounds

**Column B**

- a. decomposition reaction
- b. activity series
- c. spectator ions
- d. balanced equation
- e. double-replacement reaction
- f. catalyst
- g. combustion reaction
- h. combination reaction
- i. ionic equation
- j. single-replacement reaction

**B. Multiple Choice**

Choose the best answer and write its letter on the line.

- \_\_\_\_\_ 11. In the chemical equation  $2\text{H}_2\text{O}_2(\text{aq}) \xrightarrow{\text{MnO}_2} 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$ , the  $\text{MnO}_2$  is an
  - a. reactant.
  - b. product.
  - c. spectator ion.
  - d. catalyst.
- \_\_\_\_\_ 12. When the equation  $\text{Mg}(\text{s}) + \text{HCl}(\text{aq}) \rightarrow \text{MgCl}_2(\text{aq}) + \text{H}_2(\text{g})$  is balanced, what is the coefficient for  $\text{HCl}$ ?
  - a. 1
  - b. 2
  - c. 3
  - d. 4

# Chemistry Chapter 11 Chemical Reactions Test Answers

**Donald J. Wink,Sharon Fetzer-  
Gislason,Sheila McNicholas**

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