

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20. \\ \hline & + & \\ & 63.62 & \rightarrow 64 \end{array}$$

Chemistry Chapter 61 Assessment Answers

TD Snyder



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A Laboratory Manual of General Chemistry Leon Burr Richardson, Andrew Jackson Scarlett, 1929 Pharmaceutical Chemistry E-Book David G. Watson, 2011-02-09 This new book from the editor of the highly successful *Pharmaceutical Analysis* sets out to define the area of pharmaceutical chemistry as distinct from medicinal chemistry. It focuses less on prototypes of drugs that perhaps never came to market and more on the drugs currently in use. The emphasis in the book is on the physicochemical properties of drug molecules and in so far as they are known the way that these properties govern the interaction of the drug with its target. Important physicochemical properties include pKa and partition coefficient and the properties of the structural elements within the drug which provide interactions with the target via a range of intermolecular forces. The last fifteen years has seen a great advance in the knowledge of protein structures and a strong emphasis is given to the interaction of drugs with proteins which shape the majority of drug mechanisms. Features: Focus on intramolecular actions. Mechanisms of action richly illustrated. Self assessment included. Comprehensive chapters on vitamins and biotechnological products. This new book from the editor of the highly successful *Pharmaceutical Analysis* sets out to define the area of pharmaceutical chemistry as distinct from medicinal chemistry. It focuses less on prototypes of drugs that perhaps never came to market and more on the drugs currently in use. The emphasis in the book is on the physicochemical properties of drug molecules and in so far as they are known the way that these properties govern the interaction of the drug with its target. Important physicochemical properties include pKa and partition coefficient and the properties of the structural elements within the drug which provide interactions with the target via a range of intermolecular forces. The last fifteen years has seen a great advance in the knowledge of protein structures and a strong emphasis is given to the interaction of drugs with proteins which shape the majority of drug mechanisms. Features: Focus on intramolecular actions. Mechanisms of action richly illustrated. Self assessment included. Comprehensive chapters on vitamins and biotechnological products. *Lehne's Pharmacology for Nursing Care - E-Book* Jacqueline Rosenjack Burchum, Laura D. Rosenthal, 2023-12-14 Selected for Doody's Core Titles 2024 in Pharmacology Master an understanding of pharmacology and apply that understanding to nursing practice. With an engaging approach that actually makes learning nursing pharmacology enjoyable. *Lehne's Pharmacology for Nursing Care* 12th Edition makes it easier to understand difficult pharmacologic principles and the clinical use of drugs. The book discusses physiology and pathophysiology for each of the major drug families using drug prototypes to simplify learning. What's more it focuses your attention on the most important content limiting discussions of drug interactions and adverse effects to those that matter most. Written by nursing educators Jacqueline Rosenjack Burchum and Laura D Rosenthal this text emphasizes the understanding of drugs and drug therapy as opposed to simply memorizing drug facts. Clear engaging writing style simplifies complex concepts making difficult pharmacology content not only understandable but actually enjoyable. Prototype Drugs approach focuses on representative agents that characterize all members of a given drug group so

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