

## CHEMISTRY COMES ALIVE

Matter – anything with mass and occupies space

- weight – pull of gravity on mass
- solid – with definite shape and volume
- liquid – with definite volume but changeable shape
- gas – with changeable volume and shape

Energy - capacity to do work or put matter into motion

- 2 types – kinetic energy – energy in action
- potential energy – stored ( inactive ) energy
- energy can be transferred from potential to kinetic

Forms of energy used by the body:

1. Chemical energy – stored in the chemical bonds
  - ATP ( energy of the body)
2. Electrical energy - result from movement of charged particles
3. Mechanical energy – energy that directly cause the movement of matter
4. Radiant or electromagnetic energy - energy that travel in waves.
  - light, infrared waves, radiowaves, ultraviolet waves, X-rays

Energy - is neither created nor destroyed

- can be converted from one form to another form,
- some energy is unused in the conversion ( lighting a bulb)

Composition of Matters: All matters are made up of elements

Elements: - unique substances that cannot be broken into simpler form by ordinary chemical means

- 2 properties of elements – physical – anything we can sense or measure
- chemical – how atoms interact (bond ) with other atoms

Atoms -unique building blocks of elements

- give each element its chemical and physical properties
- smallest particles of the element that retain the properties of the element
- atomic symbols – designated for each element.
  - usually represented by the first letter or the first 2 letter of the name of the element in English, Latin mainly and rarely in other languages
- 118 elements – 92 are naturally occurring
- 26 are produced from natural element by particle accelerator or nuclear reactor.

# Chemistry Comes Alive Study Guide Answers

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## **Table of Contents Chemistry Comes Alive Study Guide Answers**

1. Understanding the eBook Chemistry Comes Alive Study Guide Answers
  - The Rise of Digital Reading Chemistry Comes Alive Study Guide Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Comes Alive Study Guide Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Comes Alive Study Guide Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Comes Alive Study Guide Answers
  - Personalized Recommendations
  - Chemistry Comes Alive Study Guide Answers User Reviews and Ratings
  - Chemistry Comes Alive Study Guide Answers and Bestseller Lists
5. Accessing Chemistry Comes Alive Study Guide Answers Free and Paid eBooks
  - Chemistry Comes Alive Study Guide Answers Public Domain eBooks
  - Chemistry Comes Alive Study Guide Answers eBook Subscription Services
  - Chemistry Comes Alive Study Guide Answers Budget-Friendly Options
6. Navigating Chemistry Comes Alive Study Guide Answers eBook Formats

- ePub, PDF, MOBI, and More
  - Chemistry Comes Alive Study Guide Answers Compatibility with Devices
  - Chemistry Comes Alive Study Guide Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chemistry Comes Alive Study Guide Answers
    - Highlighting and Note-Taking Chemistry Comes Alive Study Guide Answers
    - Interactive Elements Chemistry Comes Alive Study Guide Answers
  8. Staying Engaged with Chemistry Comes Alive Study Guide Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chemistry Comes Alive Study Guide Answers
  9. Balancing eBooks and Physical Books Chemistry Comes Alive Study Guide Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chemistry Comes Alive Study Guide Answers
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Chemistry Comes Alive Study Guide Answers
    - Setting Reading Goals Chemistry Comes Alive Study Guide Answers
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Chemistry Comes Alive Study Guide Answers
    - Fact-Checking eBook Content of Chemistry Comes Alive Study Guide Answers
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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