

CP Chemistry: Final Exam Review Packet

- 1) In a chemical reaction, the name(s) of the material(s) that you start with are called the reactants and appear on the left side of the arrow.
- 2) In a chemical reaction, the name(s) of the material(s) that you end with are called the products and appear on the right side of the arrow.
- 3) In a chemical reaction, there is a conservation of
A) moles **B) atoms** C) molecules D) temperature
- 4) When calcium carbonate reacts with nickel (II) acetate the name of one of the new compounds formed is
A) carbonate sulfate B) calcium calcitate C) nickel (II) calcium
D) nickel (II) carbonate
- 5) The complete combustion of a hydrocarbon, such as C_6H_{14} , involves the reaction with oxygen gas to form water and which of the following:
A) CO **B) CO_2** C) C D) CHO
- 6) Which of the following reactions is a decomposition reaction?
A) $Al_2O_3 + 6Li \rightarrow 3Li_2O + 2Al$
B) $2Al(OH)_3 \rightarrow Al_2O_3 + 3H_2O$
C) $2H_2 + O_2 \rightarrow 2H_2O$
D) $CH_4 + 2O_2 \rightarrow CO_2 + 2H_2O$
- 7) The reaction of dilute hydrochloric acid and zinc will produce
A) zinc oxide and water
B) zinc chloride and water
C) zinc chloride and hydrogen
D) zinc hydride and chlorine
- 8) Aluminum nitrate and lead (II) acetate react to produce
A) $AlC_2H_3O_2$ and $PbNO_3$
B) $Al(C_2H_3O_2)_3$ and $Pb(NO_3)_2$
C) $AlC_2H_3O_2$ and $Pb(NO_3)_2$
D) $Al(C_2H_3O_2)_3$ and $PbNO_3$

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AW Rasmussen



Chemistry Cp 6 Final Review:

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