

CP CHEMISTRY STUDY GUIDE • UNIT 2

Science of Matter (Chapter 2, with sections of Chapters 5 and 10)

Unit Goals ("During this unit I will...")

1. Describe and examine the various classifications of matter, both general and specific.
2. Differentiate between physical and chemical properties, both qualitative and quantitative.
3. Develop a working knowledge of two (2) quantitative physical properties: density and specific heat capacity, and their associated mathematical formulas.
4. Evaluate the Law of Conservation of Energy both descriptively and quantitatively.

PA Academic Standards - Science & Technology

3.1.10.B. Describe concepts of models as a way to predict and understand science and technology.
• Apply mathematical models to science and technology

3.2.10.B. Apply process knowledge and organize scientific and technological phenomena in varied ways.
• Describe materials using precise quantitative and qualitative skills based on observations.

3.4.10/12.A. Explain and apply concepts about the structure and properties of matter.
• Apply knowledge of mixtures to appropriate separation techniques.
• Quantify the properties of matter (e.g., density, solubility coefficients) by applying mathematical formulas.

3.4.12.B. Apply and analyze energy sources and conversions and their relationship to heat and temperature.
• Apply appropriate thermodynamic concepts (e.g., conservation, entropy) to solve problems relating to energy and heat.

Essential Skills ("Upon completion of this unit I should be able to...")

YES

I Can!

- Explain that all matter is particulate in nature. (2.1)
- Compare and contrast the following terms: matter, elements, compounds, atoms, molecules, pure substances, and mixtures (homo- and heterogeneous) (2.1, 2.2, 2.5)
- Describe, in general terms, the common states of matter (2.3)
- Describe, and give examples of, chemical and physical properties (2.4)
- Describe, and give examples of, chemical and physical changes (2.4)
- Describe, and give examples of, various techniques that could be used to separate mixtures based on the physical properties of their constituents (2.5-2.6)

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Michael Brown

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