

AP Chemistry Final Study Guide

Unit 1: Atomic Structure and Properties (7%-9%)

- At STP, one mole of gas always occupies 22.4 L
- **Mass Spectroscopy:**
 - Atoms have different isotopes (carbon-12, carbon-14, etc) and this will affect the average atomic mass
 - The most abundant isotope will have the greatest effect
- **Coulomb's Law:**
 - The greater the charge of the nucleus, the more energy is required to ionize an electron
 - The closer an electron is to the nucleus, the more energy is required to ionize an electron
- **Electromagnetic Radiation:**
 - When atoms absorb energy, electrons jump to higher energy levels
 - When electrons drop from higher to lower energy levels, atoms give off electromagnetic radiation energy
- **Kinetic Energy of Ionized Electrons:**
 - The faster an ejected electron is going, the more kinetic energy it has
 - Electrons originally further away from the nucleus require less energy to eject, so it will have more kinetic energy
- **Electron Configuration:**
 - When entering the d-block (transition metals), the energy level drops by 1 unless you are using shorthand notation
 - Nickel: $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 4s^2$ (notice the d-block)
 - Nickel (with shorthand): $[\text{Ar}]4s^2 3d^8$
- **Law of Definite Proportions:**
 - A pure chemical compound broken down into elements always contains elements of a fixed ratio, independent of where and how it was created
- **Law of Multiple Proportions:**
 - When elements form compounds, the proportions of the elements in those chemical compounds can be expressed in small whole number ratios
- **Aufbau Principle:**
 - When building up the electron configuration of an atom, electrons are placed in orbitals, subshells, and shells in order of increasing energy

Chemistry Final Exam Concepts Study Guide

John McMurry, Susan McMurry

Chemistry Final Exam Concepts Study Guide:

Basic Concepts of Chemistry, Study Guide Leo J. Malone, 2000 The text's three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry. Designed to help readers master the principles of general chemistry. As a prep book it promotes active involvement with the material. There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills. Updated to include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems.

Learning Science Through Computer Games and Simulations National Research Council, Division of Behavioral and Social Sciences and Education, Board on Science Education, Committee on Science Learning: Computer Games, Simulations, and Education, 2011-05-12 At a time when scientific and technological competence is vital to the nation's future the weak performance of U.S. students in science reflects the uneven quality of current science education. Although young children come to school with innate curiosity and intuitive ideas about the world around them science classes rarely tap this potential. Many experts have called for a new approach to science education based on recent and ongoing research on teaching and learning. In this approach simulations and games could play a significant role by addressing many goals and mechanisms for learning science the motivation to learn science conceptual understanding science process skills understanding of the nature of science scientific discourse and argumentation and identification with science and science learning. To explore this potential Learning Science Computer Games Simulations and Education reviews the available research on learning science through interaction with digital simulations and games. It considers the potential of digital games and simulations to contribute to learning science in schools in informal out of school settings and everyday life. The book also identifies the areas in which more research and research based development is needed to fully capitalize on this potential. Learning Science will guide academic researchers developers publishers and entrepreneurs from the digital simulation and gaming community and education practitioners and policy makers toward the formation of research and development partnerships that will facilitate rich intellectual collaboration. Industry government agencies and foundations will play a significant role through start up and ongoing support to ensure that digital games and simulations will not only excite and entertain but also motivate and educate.

Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011

Crimes Against the English Language, **Study Guide for Chemistry by Steven S. Zumdahl** Martha B. Barrett, 1986

Study Guide to accompany Basic Concepts of Chemistry, 7th Edition Leo J. Malone, 2003-02-20

Work more effectively and gauge your progress along the way. This Study Guide that is designed to accompany Malone's Chemistry 7th Edition includes chapter summaries new terms self tests answers to self tests and solutions to selected problems. This easy to read introduction presents chemistry as a living relevant science. Chemistry 7th Edition encourages

critical thinking and helps readers overcome the math difficulties that often prevent them from developing a full understanding of the subject. This new seventh edition builds on its core strengths of pedagogy driving the connections between ideas mathematics in context not just an appendix and an extensive problem solving emphasis with an updated design and more molecular art. In addition the seventh edition expands its applications and online options. One of the briefest books in the market it still provides sufficient depth for the basic concepts of chemistry.

Robinson Chemistry Study Guide Robinson, 1992 *Study Guide and Solutions Manual for Fundamentals of Organic Chemistry* Susan McMurry, 1994

Following a brief review of structure and bonding organic molecules and functional groups are presented as early as possible. The text is organized primarily by functional group beginning with simple alkanes and moving toward more complex compounds. Emphasis is placed on the fundamental mechanistic similarities of organic reactions. McMurry's thorough revision continues to present the solid content necessary for this course without sacrifice of important subjects and pedagogical tools. Text and reaction summaries full problem sets and outstanding artwork are just some of the features in the Third Edition usually found in a full year book. McMurry's clear well written explanations remain a highlight of the book.

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Study Guide for Chang's Chemistry** Kenneth W. Watkins, 1988 *Students Study Guide* Fredeen, Pearson, 2003-07 **Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry** McMurry, Susan McMurry, 1990 [Acs General Chemistry Study Guide](#)

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