

## CHEMISTRY FINAL EXAM REVIEW GUIDE 2012 Scott

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### Multiple Choice

*Identify the choice that best completes the statement or answers the question.*

- 1) Which of the following observations is quantitative?
  - a. The liquid turns blue litmus paper red.
  - b. The liquid boils at 100°C.
  - c. The liquid tastes bitter.
  - d. The liquid is cloudy.
  
- 2) The symbols for units of length in order from largest to smallest are
  - a. m, cm, mm, km.
  - b. mm, m, cm, km.
  - c. km, mm, cm, m.
  - d. km, m, cm, mm.
  
- 3) A volume of 1 milliliter is equivalent to
  - a. 1 cubic centimeter.
  - b. 1 gram.
  - c. 1 liter.
  - d.  $10^{-3}$  cubic decimeters.
  
- 4) Which pair of quantities determines the density of a material?
  - a. mass and weight
  - b. volume and weight
  - c. volume and concentration
  - d. volume and mass
  
- 5) A sample of gold has a mass of 96.5 g and a volume of 5.00 cm<sup>3</sup>. The density of gold is
  - a. 0.0518 g/cm<sup>3</sup>.
  - b. 19.3 g/cm<sup>3</sup>.
  - c. 101.5 g/cm<sup>3</sup>.
  - d. 48.3 g/cm<sup>3</sup>.
  
- 6) The number of grams equal to 0.5 kg is
  - a. 0.0005.
  - b. 0.005.
  - c. 500.
  - d. 5000.
  
- 7) If 1 inch equals 2.54 cm, how many centimeters equal 1 yard?
  - a. 0.0706 cm
  - b. 14.2 cm
  - c. 30.5 cm
  - d. 91.4 cm
  
- 8) A chemist obtained the values 5.2246 g, 5.2353 g, and 5.2501 g for the mass of a sample. Without knowing the true mass of the sample, it can be said that these values have
  - a. good precision.
  - b. good accuracy.
  - c. poor precision.
  - d. poor accuracy.
  
- 9) The measurement 0.035550 g rounded off to two significant figures would be
  - a. 0.03 g.
  - b. 0.35 g.
  - c. 0.036 g.
  - d.  $3.5 \times 10^2$  g.
  
- 10) When 64.4 is divided by 2.00, the correct number of significant figures in the result is

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**George Horemis**

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