


Syllabus For Chemistry Practical exam (Strictly according to latest guidelines of CBSE)

1. Volumetric Analysis (Titration) (8)
 - (i) To prepare standard solution of Mohr's salt (M/10 strength) and determine the concentration of given potassium permanganate solution in g/L.
 - (ii) To prepare standard solution of oxalic acid (M/10 strength) and determine the concentration of given potassium permanganate solution in g/L.
2. Qualitative analysis of inorganic salt. (8)

Group	Basic radical (cation)	Group	Acidic radical (anion)
Zero	NH_4^+	First	S^{2-} , CH_3COO^- , CO_3^{2-} , SO_3^{2-} , NO_2^-
First	Pb^{+2} , Ag^+ , Hg_2^{+2}		
Second	Hg^{+2} , Bi^{+3} , Pb^{+2} , Cu^{+2} , Cd^{+2}	Second	Cl^- , Br^- , I^- , NO_3^- , $C_2O_4^{2-}$
Third	Al^{+3} , Fe^{+3} , Cr^{+3}	Third	SO_4^{2-} , PO_4^{3-}
Fourth	Zn^{+2} , Ni^{+2} , Co^{+2} , Mn^{+2}		
Fifth	Ba^{+2} , Sr^{+2} , Ca^{+2}		
Sixth	Mg^{+2}		

3. Content based Experiment / Test of functional group in the given organic salt. (6)

Functional group	Structural formulae.
Carboxylic acid	$-COOH$
Phenol	
Alcohol	$-OH$
Aldehyde	$-CHO$
Ketone	$-CO-$
Ester	$-COOR$
Amines	$-NH_2$
amide	$-CONH_2$

4. Investigatory project .
 - (i) Study of the presence of oxalate ion in guava fruit at different stages of ripening.
 - (ii) Study of casein quantity present in different sample of milk.
 - (iii) Preparation of soyabean milk and its comparison with natural milk.
 - (iv) Potassium bisulphate as a food preservative.
 - (v) Study of digestion of starch by salivary amylase.
 - (vi) Comparative study of the rate of fermentation of various food material.
 - (vii) To extract the essential oils present in the naturally occurring material.
 - (viii) Study of adulterants in food – stuffs.
5. Practical record and viva – voce

Chemistry Practical Manual Class 12 Cbse 2013

**Dr. N. K. Verma, B. K. Vermani, Dr.
Neera Verma**



Chemistry Practical Manual Class 12 Cbse 2013:

Disha Combo (4 Books) 13 Years CBSE Class 12 Solved Physics, Chemistry, Mathematics & English Core Previous Year-wise Question Papers (2013 - 2025) 4th Edn | PYQs | 2026 CBSE Question Bank Class 12 , The latest updated 5th Edition of the books 13 Year CBSE Board Physics Chemistry Mathematics English Core Class 12 YEAR WISE Solved Papers 2013 2025 powered with Concept Notes is a must have book for aspirants who are looking for better score in CBSE exams The Books contain the Past 13 Years Authentic Solved papers of CBSE Board Class 12 including 1 set each of All India Delhi 2025 Exam In all the Books contain 25 Papers including the 2021 CBSE Sample Paper This paper has been included as this year the Board exams were cancelled The USP of the book is the inclusion of Concept Notes highlighting Tips Tricks Alternate solutions Points to Remember in various solutions The Notes will help the students in further revision of syllabus Trend Analysis of of last 6 Years is provided to understand Question trend Errorless Solutions with step by step marking scheme on the lines of CBSE Board and written in a way that any student can understand easily **Oswaal CDS**

(Combined Defence Services) 14 Solved Papers Year-wise 2018-2024 (II) | General Knowledge | For 2025 Exam

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would like to thank our authors editors and reviewers Special thanks to our students who send us suggestions and constantly help improve our books To stay true to our motto of Learning Made Simple we constantly strive to present information in ways that are easy to understand as well as remember

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familiarise me with exam patterns Some benefits of studying from Oswaal CDS check 14 Previous year solved question papers 100% updated with Fully Solved Paper of September 2024 II Concept Clarity with detailed explanations of 2018 to 2024 Papers Extensive Practice with 1600 Questions and Two Sample Question Papers Crisp Revision with Mind Maps Expert Tips helps you get expert knowledge master crack CDS in first attempt Exam insights with Previous Years 2024 2019 Trend Analysis empowering students to be 100% exam ready Our Heartfelt Gratitude Finally we would like to thank our authors editors and reviewers Special thanks to our students who send us suggestions and constantly help improve our books To stay true to our motto of Learning Made Simple we constantly strive to present information in ways that are easy to understand as well as remember

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CUET (UG) 20 Combined Mock Test Papers Science (PCM) Stream (English, Physics, Chemistry, Mathematics, General Test) For 2025 Exam Oswaal Editorial Board,2024-09-05 The Central Universities Entrance Test CUET UG has emerged as a crucial examination for students aspiring to secure admission into prestigious universities across India With the competition intensifying each year thorough preparation and strategic practice have become essential for success In 2024 The Common University Entrance Test CUET had conducted the assessment test for admission to UG programme across 44 Central

Universities at 354 centres in India and at 25 centres abroad In 2024 nearly 15 lakh candidates registered for the CUET UG Exam Oswaal Combined Mock Test Papers have been meticulously designed to simulate the syllabus and actual exam pattern issued by the NTA ensuring that students can practice in an environment similar to the one they will face on the day of the examination Each test paper covers a wide range of topics carefully aligned with the CUET syllabus helping students assess their knowledge hone their problem solving skills and improve their time management in the CBT test A Hybrid mode Pen Paper CBT question paper is divided into four sections Sections IA IB contain language specific questions Section II comprises domain related questions and Section III contains questions on general topics All questions are MCQ based This book of Combined Mock Test Papers for CUET UG aims to provide a comprehensive and well structured resource for students preparing in various streams including Commerce English Accountancy Business Studies Economics and General Test Humanities English History Geography Political Science and General Test Science PCB PCM English Physics Chemistry Biology Mathematics and General Test Here's how this book benefits you 100% updated with the Latest Solved Paper 2024 for unmatched accuracy relevance 20 Mock test papers 4 sets in each subject for extensive practice 1200 Questions approx for comprehensive coverage Answer key with explanations to ensure in depth concept clarity OMR Sheet simulates the real exam experience We believe that consistent practice and familiarity with the exam format are key to excelling in CUET UG By working through these Mock Test Papers students will not only build a strong foundation in their respective subjects but also develop the confidence needed to tackle the exam with ease We hope this book serves as a valuable companion in your preparation journey We wish all aspiring students the very best of luck in their endeavours and future academic pursuits

Comprehensive Practical Chemistry XII Dr. N . K. Verma, B. K. Vermani, Dr. Neera Verma, K . K. Rehani, 2011-11

Chemistry Lab Manual Class XII Follows the Latest CBSE Syllabus and Other State Board Following the CBSE

Curriculum. Edugorilla Prep Experts, 2022-07-05 Chemistry is all about conducting various experiments and having a deep study about them This deep analysis and study cannot solely be done with theories and practical experiments are necessary for better skill development of the students Considering this every year CBSE conducts the chemistry practicals in addition to the theory examinations This class 12 chemistry lab manual contains all the essential information that the students will require while commencing the practicals CBSE class 12th chemistry lab manual has been prepared strictly according to CBSE guidelines which follows the NCERT and other State Board curriculum All the experiments in this lab manual have been given in a simple lucid and interactive manner The theory given with each experiment gives complete knowledge of each term concept and definition in order to provide a compact knowledge to the students at one place Viva Voce questions are given with each experiment to test students understanding of related experiments Mr Kaushalesh Dwivedi is the author of this complete lab activity book for class 12 chemistry All the lab based experiments have well labeled diagrams which clearly demonstrate the details of the apparatus that are used the correct way of handling Laboratory apparatus and perform

the experiments methodically The basic purpose of this lab manual is to provide the students with appropriate guidance so that they can easily understand carry out and document the experiments in the Science Laboratory The Lab Manual comes with practical lessons followed by Lab experiments after each chapter projects and Notebook for jotting down all the essential information The aim of this Science Lab Manual is to help class 12 students to prepare and understand the complex chemistry concepts through experimental approach by utilizing pedagogically enhanced content of the lab manual and with a compact knowledge base for improving their analytical experimental skills and learning abilities

Practical/Laboratory Manual Chemistry Class - XI Er. Meera Goyal, 2021-05-29

1 Basic Laboratory Techniques

- 1 To cut a glass tube or glass rod
- 2 To bend the glass rod at an angle
- 3 To draw a glass jet from a glass tube
- 4 To bore a cork and fit a glass tube into it

Viva Voce

2 Characterisation and Purification of Chemical Substances

- 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique
- Viva Voce
- 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method
- Viva Voce
- 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample
- 4 To prepare the pure crystals of copper sulphate from the given crude sample
- 5 To prepare pure crystals of benzoic acid from a given impure sample

Viva Voce

3 Measurement of pH Values

- 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper
- 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration
- 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper
- 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH
- 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+

Viva Voce

4 Chemical Equilibrium

- 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions
- 2 To study the shift in equilibrium between $Co(H_2O)_6^{2+}$ and Cl^- ions by changing the concentrations of either of the ions

Viva Voce

5 Quantitative Analysis

- 1 To prepare M 10 oxalic acid solution by direct weighing method
- 2 To prepare M 10 solution of sodium carbonate by direct weighing method
- 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid
- 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution

Viva Voce

6 Qualitative Analysis

Analysis of Anions

Analysis of Cations

Viva Voce

7 Detection of Elements in Organic Compounds

- 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test
- 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test

Viva Voce

INVESTIGATORY PROJECTS

- 1 Checking of Bacterial Contamination in Water
- 1 To check the bacterial contamination in drinking water by testing sulphide ions
- Viva Voce
- 2 Methods of Water Purification
- 1 To purify water from suspended impurities by using sedimentation
- 2 To purify water by boiling
- 3 To purify water by distillation method
- 4 To purify water by reverse osmosis technique
- 5 To purify water by GAC method
- 6 To purify water by bleach

treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method Viva Voce 3 Water Analysis 1 To test the hardness of different water samples Viva Voce 4 Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids lViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce

Practical/Laboratory Manual Chemistry Class XII based on NCERT guidelines by Dr. S. C. Rastogi, Er. Meera Goyal Dr. S. C. Rastogi ,Er. Meera Goyal,2020-06-22 A Surface Chemistry 1 To prepare colloidal solution sol of starch 2 To prepare a colloidal solution of egg albumin 3 To prepare colloidal solution of gum 4 To prepare colloidal solution of aluminium hydroxide $Al(OH)_3$ 5 To prepare colloidal solution of ferric hydroxide $Fe(OH)_3$ 6 To prepare colloidal solution of arsenious sulphide As_2S_3 7 To purify a freshly prepared sol by dialysis 8 To compare the effectiveness of different common oils Castor oil cotton seed oil coconut oil kerosene oil mustard oil in forming emulsions Viva Voce B Chemical Kinetics 1 To study the effect of concentration on the rate of reaction between sodium thiosulphate and hydrochloric acid 2 To study the effect of temperature on the rate of reaction between sodium thiosulphate and hydrochloric acid 3 To study the rate of reaction of iodide ions with hydrogen peroxide at different concentrations of iodide ions 4 To study the rate of reaction between potassium iodate KIO_3 and sodium sulphite Na_2SO_3 using starch solution as indicatorl Viva Voce C Thermochemistry 1 Determine the enthalpy of dis solution of copper sulphate $CuSO_4 \cdot 5H_2O$ in water at Room temperature 2 To determine the enthalpy of neutralization of the reaction between HCl and $NaOH$ 3 To determine enthalpy change during the interaction between acetone and chloroform Viva Voce D Electrochemistry 1 To study the variation of cell potential in $Zn|Zn^{2+}||Cu^{2+}|Cu$ with change in concentration of electrolytes $CuSO_4$ or $ZnSO_4$ at room temperature Viva Voce E Chromatography 1 To separate the coloured components pigment present in the given extract of leaves and flowers by ascending paper chromatography and find their R_f values 2 To separate the coloured components present in the mixture of red and blue inks by ascending paper chromatography and find their R_f values 3 To separate Co^{2+} and Ni^{2+} ions present in the given mixture by using ascending paper chromatography and determine their R_f values Viva Voce F Preparation of Inorganic Compounds 1 Preparation of double salt of ferrous ammonium sulphate Mohr s salt from ferrous sulphate and ammonium sulphate 2 To prepare a pure sample of potash alum fitkari 3 Preparation of crystals of potassium ferric oxalate or pottasium trioxlato ferrate III Viva Voce G Preparation of Organic Compounds 1 Preparation of iodoform from ethyl alcohol or acetone 2 Preparation of acetanilide in laboratory 3 Preparation of b Naphthol aniline dye 4 To prepare a pure sample of dibenzalacetone 5 To prepare a pure sample of p nitro acetanilide Viva Voce H Tests for the Functional

Groups Present in Organic Compounds Viva Voce I Study of Carbohydrates Fats and Proteins 1 To study simple reactions of carbohydrate 2 To study simple reactions of fats 3 To study simple reactions of proteins 4 To investigate presence of carbohydrates fats and proteins in food stuffs Viva Voce J Volumetric Analysis 1 To prepare 250 ml of M 10 solution of oxalic acid 2 To prepare 250 ml of M 10 solution of ferrous ammonium sulphate 3 Prepare M 20 solution of oxalic acid with its help find out the molarity and strength of the given solution of potassium permanganate 4 Prepare M 20 solution of Mohr's salt using this solution determine the molarity and strength of potassium permanganate solution Viva Voce K Qualitative Analysis Viva Voce INVESTIGATORY PROJECTS 1 To study the presence of oxalate ions in guava fruit at different stages of ripening 2 To study the quantity of casein present in different samples of milk 3 Preparation of soyabean milk and its comparison with natural milk with respect to curd formation effect of temperature etc 4 To study the effect of potassium bisulphite as food preservative at various concentrations 5 To study the digestion of starch by salivary amylase and the effect of pH and temperature on it 6 To study and compare the rate of fermentation of the following materials wheat flour gram flour potato juice and carrot juice 7 To extract essential oils present in saunf aniseed ajwain coriander illaichi cardamom 8 To detect the presence of adulteration in fat oil and butter 9 To investigate the presence of NO_2 in brinjal

Comprehensive Practical Chemistry (Abhilekhan) XII Dr. N.K. Verma, Avinash Ojha, 2008-05-01 **Practical/Laboratory Manual Chemistry**

Class XI based on NCERT guidelines by Dr. S. C. Rastogi & Er. Meera Goyal Dr. S. C. Rastogi, Er. Meera

Goyal, 2020-06-23 An Excellent Book in Accordance with the latest syllabus for Class 11 Prescribed by CBSE NCERT and Adopted by Various State Education Boards A Basic Laboratory Techniques 1 To cut a glass tube or glass rod 2 To bend the glass rod at an angle 3 To draw a glass jet from a glass tube 4 To bore a cork and fit a glass tube into it B Characterisation and Purification of Chemical Substances 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method 3 To prepare crystals of pure potash alum $\text{K}_2\text{SO}_4 \cdot \text{Al}_2(\text{SO}_4)_3 \cdot 24\text{H}_2\text{O}$ from the given impure sample 4 To prepare the pure crystals of copper sulphate from the given crude sample 5 To prepare pure crystals of benzoic acid from a given impure sample C Measurement of pH Values 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+ D Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2 To study the shift in equilibrium between $\text{Co}(\text{H}_2\text{O})_6^{2+}$ and Cl^- ions by changing the concentrations of either of the ions E Quantitative Analysis 1 To prepare M 10 oxalic acid solution by direct weighing method 2 To prepare M 10 solution of sodium

carbonate by direct weighing method 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution F Qualitative Analysis 1 Analysis of Anions 2 Analysis of Cations G Detection of Elements in Organic Compounds 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne s test 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne s test INVESTIGATORY PROJECTS A Checking of Bacterial Contamination in Water 1 To check the bacterial contamination in drinking water by testing sulphide ions B Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 To purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method C Water Analysis 1 To test the hardness of different water samples D Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap E Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper F Analysis of Fruits and Vegetable Juices 1 To analyse the fruit and vegetable juices for the constituent present in them G Rate of Evaporation 1 To study the rate of evaporation of different liquids H Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Log Antilog Table

Chemistry Practical Manual Withers,1971 Year 12 Chemistry Practical Workbook Graeme Abbott,Anne Hume,Bev Cooper,1998-07-01 *Chemistry Lab Manual Class XI | Follows the Latest CBSE Syllabus and Other State Board Following the CBSE Curriculam.* Edugorilla Prep Experts,2022-07-05 Theories are incomplete without experimentation So in order to enhance the practical skills of students the CBSE conducts the practical exams for class 11 chemistry along with theoretical exams to test the scientific aptitude of students CBSE Class 11th chemistry Lab Manual has been prepared strictly according to CBSE guidelines which follows the NCERT and other state board curriculum All the experiments in this Lab Manual have been given in a simple lucid and interactive manner The theory given with each experiment gives complete knowledge of each term concept and definition in order to provide a compact knowledge to the students at one place Viva Voce questions are given with each experiment to test students understanding of related experiments Mr Kaushalesh Dwivedi is the author of this complete lab activity book for class 11 chemistry All the lab based experiments have well labeled diagrams which clearly demonstrate the details of the apparatus that are used the correct way of handling Laboratory apparatus and perform the experiments methodically The basic purpose of this Lab Manual is to provide the students with appropriate guidance so that they can easily understand carry out and document the experiments in the Science Laboratory The Lab Manual comes with practical lessons followed by lab experiments after each chapter projects and Notebook for jotting down all the essential

information this develops a better understanding about the concept behind each class 11 chemistry theory there is an experiment which brings authentic and easy to retain conclusions through its observations and which is verified on a real time basis The aim of this Science Lab Manual is to help class 11 students to prepare and understand the complex chemistry concepts through experimental approach by utilizing pedagogically enhanced content of the Lab Manual and with a compact knowledge base for improving their analytical experimental skills and learning abilities **Practical Chemistry (For B.Sc.**

I, II and III Year Students) Pandey O.P. & Bajpai D.N. & Giri S.,1972 FOR B Sc I II III YEAR STUDENTS

Comprehensive Practical Chemistry XI Dr. N. K. Verma,B. K. Vermani,Dr. Neera Verma,2010-02

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