

Chapter 10 Review: Chemical Quantities & Calculations

(molar – mole & volume calculations, % composition, empirical & molecular formulas)

For the test, be sure that you are familiar, and able to complete, the problems below as well as the problems found on the following worksheets: "Molecular Weight and Mole Calculations," "Percent Comp. & Chemical Formula Calculations" and "Chapter 10 Sample Problems." Check the class website for answers to the review worksheet.

1. The air around you is made up of 20.95% oxygen. How many oxygen molecules are in 75.0 grams of air? The density of air is 1.19 g/L. *assume STP.*
2. How much helium (grams) would be needed to inflate a balloon to a volume of 5.50L? *assume STP.*
3. The molecular formula of an antibacterial drug is $C_{17}H_{18}FN_2O_5$. How many fluorine atoms are in 150mg of this drug?
4. A molecule is made up of 40.0% carbon, 6.67% hydrogen, and an unknown amount of oxygen. What is the molecular formula of the molecule if its molar mass is 90.0g/mole?
5. What is the empirical formula for a compound made up of 71.72% chlorine, 16.16% oxygen and 12.12% carbon? Calculate the molecular formula for the compound if it has a molar mass of 296.7g/mol.
6. How many molecules are in 8.42 Kilograms of C_2H_6 ?

Chapter 10 Chemical Quantities Guided Practice

Problems Answers

Julie R. Frentrup

Chapter 10 Chemical Quantities Guided Practice Problems Answers:

Study Guide/Selected Solutions Manual Julie R. Frentrup,2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book

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In a world consumed by screens and the ceaseless chatter of fast transmission, the melodic elegance and mental symphony developed by the prepared term usually diminish into the backdrop, eclipsed by the relentless sound and disturbances that permeate our lives. But, nestled within the pages of **Chapter 10 Chemical Quantities Guided Practice Problems Answers** a marvelous literary treasure filled with organic feelings, lies an immersive symphony waiting to be embraced. Crafted by an outstanding musician of language, that fascinating masterpiece conducts viewers on an emotional trip, skillfully unraveling the concealed songs and profound impact resonating within each carefully constructed phrase. Within the depths with this moving assessment, we shall discover the book is key harmonies, analyze their enthralling writing type, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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