

Solutions

Reviewing Vocabulary

Match the definition in Column A with the term in Column B.

Column A

- _____ 1. The diffusion of solvent particles across a semipermeable membrane from areas of lower solute concentration to areas of higher solute concentration
- _____ 2. A mixture with particles that settle out if undisturbed
- _____ 3. The erratic movement of colloid particles
- _____ 4. The amount of additional pressure caused by water molecules moving into a solution
- _____ 5. A measure of how much solute is dissolved in a specific amount of solvent or solution
- _____ 6. The overall energy change that occurs when a solution forms
- _____ 7. A heterogeneous mixture of intermediate size particles
- _____ 8. The process of surrounding solute particles with solvent particles to form a solution
- _____ 9. The ratio of the number of moles of solute in solution to the total number of moles of solute and solvent
- _____ 10. The scattering of light by dispersed colloid particles
- _____ 11. The statement that the solubility of a gas in a liquid is directly proportional to the pressure of the gas above the liquid

Column B

- a. Brownian motion
- b. colloid
- c. concentration
- d. heat of solution
- e. Henry's law
- f. mole fraction
- g. osmosis
- h. osmotic pressure
- i. solvation
- j. Tyndall effect
- k. suspension

Describe each pair of related terms.

12. soluble, insoluble

13. miscible, immiscible

14. molarity, molality

Chapter 15 Chemistry Solutions Study Guide

Paul A. Bartlett, Judith G. Koch

Chapter 15 Chemistry Solutions Study Guide:

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