

Name Valentina Da Silva Period **Chapter 3: The Molecules of Life***Guided Reading Activities***Chapter Content: Organic Compounds**

Complete the following questions as you read the third chapter content—Organic Compounds:

- The element carbon is essential to an organic compound.
- Is this molecule an organic compound? Briefly explain your answer.

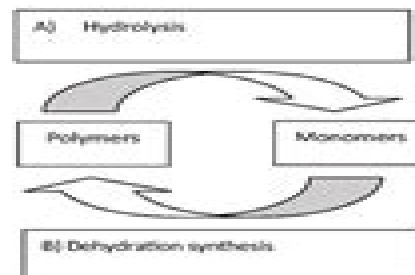


Yes, since there are carbon and hydrogen elements within the molecule.

- The oxygen atom in ethyl alcohol (the molecule pictured above) forms a polar covalent bond with the hydrogen atom. This causes the hydroxyl functional group ($-\text{OH}$) to have polarity (it is polar). Based on this, would you expect ethyl alcohol to readily interact with water? Briefly explain your answer. *Hint—Think back to the end of chapter two.*

A functional group is a hydroxyl group. Ethanol interacts with water because the hydroxyl group imparts polarity to the ethanol, which permits it to be attracted to water.

- Complete the following diagram using the following terms: dehydration reaction and hydrolysis.



Chapter 3 Guided Reading Answers

G Thomas



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