

Chemistry Finals Study Guide

Chapter 1 & 2 - Review of Basic Concepts

- **Describe the 3 states of matter.** The three states of matter are solid, liquid, and gas.
 - **Solid:** When atoms in a material are moving slow that they stick together and barely move, creating a hard object classified as a solid
 - **Liquid:** When atoms are moving slower than a gas state but faster than the solid state so they flow and are able to take the shape of their container. They have more energy than solid state atoms.
 - **Gas:** When atoms are moving faster than the other two states of matter (solid & liquid) that they are no longer pulled towards each other and roam freely without boundaries if not kept in a container.
- **Describe the difference between elements and compounds, mixtures.**
 - **Elements:** materials that are only made out of the same type of atoms and cannot be changed into simpler substances by a chemical change (ex: O₂)
 - **Compounds:** materials that are a mix of two or more element in a fixed proportion
 - **Mixtures:** a mixture of 2 or more pure substances.
 - **Heterogeneous Mixture:** can often be separated by filtration.
 - **Homogeneous Mixture:** can be separated by distillation, crystallization, and chromatography
- **Convert between Celsius and Kelvin.**
 - **Celsius to Kelvin:** $K = ({}^{\circ}C + 273.15)$
 - **Kelvin to Celsius:** ${}^{\circ}C = (K - 273.15)$
- **Round measured and calculated values to the correct number of significant digits.**
 - Significant Digit Rules:  [CHEM] Final 1 - Steps & Processes
- **Use dimensional analysis to solve multi step equations, showing how units cancel**
 -  [CHEM] Final 1 - Steps & Processes
- **What is the difference between precision and accuracy?**
 - **Precision:** Acquiring the same result again and again
 - **Accuracy:** A result that is close to the accepted value/actual answer

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NES Chemistry Exam Prep Team

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1. Understanding the eBook Chem Final Study Guide
 - The Rise of Digital Reading Chem Final Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chem Final Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chem Final Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chem Final Study Guide
 - Personalized Recommendations
 - Chem Final Study Guide User Reviews and Ratings
 - Chem Final Study Guide and Bestseller Lists
5. Accessing Chem Final Study Guide Free and Paid eBooks
 - Chem Final Study Guide Public Domain eBooks
 - Chem Final Study Guide eBook Subscription Services
 - Chem Final Study Guide Budget-Friendly Options

6. Navigating Chem Final Study Guide eBook Formats
 - ePUB, PDF, MOBI, and More
 - Chem Final Study Guide Compatibility with Devices
 - Chem Final Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chem Final Study Guide
 - Highlighting and Note-Taking Chem Final Study Guide
 - Interactive Elements Chem Final Study Guide
8. Staying Engaged with Chem Final Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chem Final Study Guide
9. Balancing eBooks and Physical Books Chem Final Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chem Final Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chem Final Study Guide
 - Setting Reading Goals Chem Final Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chem Final Study Guide
 - Fact-Checking eBook Content of Chem Final Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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- Interactive and Gamified eBooks

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