

Name \_\_\_\_\_

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For each of the atoms below, draw the proper dot diagram. Each atom has lone pairs that will pair with its own.

# Electron Dot Diagrams: Atoms and Ions

1.	2.	3.	4.	5.	6.
7.	8.	9.	10.	11.	12.
13.	14.	15.	16.	17.	18.
19.	20. Complete electron dot diagram for carbon below:		21. Copy and finish off the electron dot diagram on a piece of paper.		

The ions in below show in the proper electron dot diagrams.

22. Chloride ion, Cl <sup>-</sup> 	23. Sodium ion, Na <sup>+</sup> 	24. Hydrogen ion, H <sup>+</sup> 	25. Hydrogen ion, H <sup>+</sup> 
26. Aluminum ion, Al <sup>3+</sup> 	27. Nitride ion, N <sup>3-</sup> 	28. Oxide ion, O <sup>2-</sup> 	29. Calcium ion, Ca <sup>2+</sup> 

# Chem Guide Answer Key

**Paul A. Bartlett, Judith G. Koch**



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