

HSC Chemistry Notes:

- Module 5: Equilibrium and Acid Reactions
- Module 6: Acid and Base Reactions
- Module 7: Organic Chemistry
- Module 8: Applying Chemical Ideas

Module 5: Equilibrium and Acid Reactions

Static and Dynamic Equilibrium

Reversibility of chemical reactions:

Formation of hydrated and dehydrated cobalt (II) chloride:

Dehydrated cobalt (II) chloride + water \rightleftharpoons hydrated cobalt (II) chloride



- When the hydrated cobalt (II) chloride is heated, it becomes blue again as the water evaporated.

Reaction of Iron (III) nitrate and potassium thiocyanate:

Iron (III) nitrate + potassium thiocyanate \rightleftharpoons iron (III) thiocyanate



- When $(\text{Fe}(\text{NO}_3)_3)$ and (KSCN) are mixed, they form a red-coloured ion, iron (III) thiocyanate
- Is a reversible reaction and forms an equilibrium

Burning Magnesium:



- An irreversible combustion reaction, releasing energy into the system, producing white magnesium oxide.

Burning Steel Wool:



- An irreversible reaction and is a combustion reaction.
- As the steel wool caught on fire and went red with sparks flying out.

Chemistry Notes Hsc

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HSC Chemistry Notes, 1984 **HSC Chemistry Complete Course Notes**, 2016-06-01 **Lecture Notes in Chemistry I. (Harvard College.)** Arthur Prescott Lothrop, Archibald Livingston Hodges, 1882 **HSC Year 12 Chemistry Notes (2019)**, 2018 **HSC Year 12 Chemistry Complete Course Notes (2022)**, 2022-03 **Lecture/tutorial Notes and Problems for an Extension Course in H.S.C. Chemistry**, 1981 HSC Year 11 Chemistry Complete Course Notes (2022), 2022-03 Atar Notes Hsc Chemistry Year 12 Topic Tests (2022-2024). Atar, 2023 **Redirecting Science** Finn Aaserud, 2003-01-30 This volume is an important study for understanding the complex interconnections between basic science and its sources of economic support in the period between the two world wars The focus of the study is on the Institute for Theoretical Physics later renamed the Niels Bohr Institute at Copenhagen University and the role of its director the eminent Danish physicist Niels Bohr in the funding and administration of the Institute Under Bohr's direction the Copenhagen Institute was a central workplace in the development and the formulation of quantum mechanics in the 1920s and later became an important center for nuclear research in the 1930s Dr Aaserud brings together the scholarship on the internal origins and development of nuclear physics in the 1930s with descriptions of the concurrent changes in private support for international basic science particularly as represented by Rockefeller Foundation philanthropy In the process the book places the emergence of nuclear physics in a larger historical context This book will appeal to historians of science physicists and advanced students in these areas **Excel Chemistry** C. M. Roebuck, 1995 Excel HSC Chemistry Jim Stamell, 2011 ISBN 9781741252996 AUTHOR Jim Stamell RRP 39 95 PAGES 428 pp SPECIFICATION Softcover perfect bound 280 mm x 210 mm STATUS New edition PUBLICATION DATE April 2008 The EXCEL HSC Chemistry guide is directly linked to the syllabus with every single dot point of the HSC Chemistry syllabus appearing in the margin of the book You can write in the guide so your study is focused and your notes are structured This guide comes in a brand new format that makes even better use of your study time up to date coverage of the core topics plus 3 Option topics Industrial Chemistry Shipwrecks Corrosion and Conservation and Forensic Chemistry this guide is organised just like the HSC syllabus so the students learn to section the theoretical part is under routine headings and the students section the practical part is under headings like First hand Second hand and Investigations and Problem Solving %this way you will be able to see at a glance what the theoretical and practical work is all main headings in each chapter 1 1 2 1 etc are directly from the syllabus word for word %this way you can easily match the Excel guide to the syllabus an alphabetical list of all the key definitions and concepts you should know from each chapter %an efficient way of learning all the definitions in one go chapter syllabus checklist with every single dot point listed in checklist form for each chapter %a fantastic way of testing that you know all the work hundreds of key concept questions with answers %questions that test you recall of knowledge in each chapter HSC type questions for every section in each chapter with clock icons to tell you how much time you will have to answer the

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