

Chemistry Study Guide

Periodic Table of the Elements

¹ ₁ H																	² ₂ He																														
³ ₃ Li	⁴ ₄ Be																	⁵ ₅ B	⁶ ₆ C	⁷ ₇ N	⁸ ₈ O	⁹ ₉ F	¹⁰ ₁₀ Ne																								
¹¹ ₁₁ Na	¹² ₁₂ Mg																	¹³ ₁₃ Al	¹⁴ ₁₄ Si	¹⁵ ₁₅ P	¹⁶ ₁₆ S	¹⁷ ₁₇ Cl	¹⁸ ₁₈ Ar																								
¹⁹ ₁₉ K	²⁰ ₂₀ Ca	²¹ ₂₁ Sc	²² ₂₂ Ti	²³ ₂₃ V	²⁴ ₂₄ Cr	²⁵ ₂₅ Mn	²⁶ ₂₆ Fe	²⁷ ₂₇ Co	²⁸ ₂₈ Ni	²⁹ ₂₉ Cu	³⁰ ₃₀ Zn	³¹ ₃₁ Ga	³² ₃₂ Ge	³³ ₃₃ As	³⁴ ₃₄ Se	³⁵ ₃₅ Br	³⁶ ₃₆ Kr																														
³⁷ ₃₇ Rb	³⁸ ₃₈ Sr	³⁹ ₃₉ Y	⁴⁰ ₄₀ Zr	⁴¹ ₄₁ Nb	⁴² ₄₂ Mo	⁴³ ₄₃ Tc	⁴⁴ ₄₄ Ru	⁴⁵ ₄₅ Rh	⁴⁶ ₄₆ Pd	⁴⁷ ₄₇ Ag	⁴⁸ ₄₈ Cd	⁴⁹ ₄₉ In	⁵⁰ ₅₀ Sn	⁵¹ ₅₁ Sb	⁵² ₅₂ Te	⁵³ ₅₃ I	⁵⁴ ₅₄ Xe																														
⁵⁵ ₅₅ Cs	⁵⁶ ₅₆ Ba			⁷² ₇₂ Hf	⁷³ ₇₃ Ta	⁷⁴ ₇₄ W	⁷⁵ ₇₅ Re	⁷⁶ ₇₆ Os	⁷⁷ ₇₇ Ir	⁷⁸ ₇₈ Pt	⁷⁹ ₇₉ Au	⁸⁰ ₈₀ Hg	⁸¹ ₈₁ Tl	⁸² ₈₂ Pb	⁸³ ₈₃ Bi	⁸⁴ ₈₄ Po	⁸⁵ ₈₅ At	⁸⁶ ₈₆ Rn																													
⁸⁷ ₈₇ Fr	⁸⁸ ₈₈ Ra			¹⁰² ₁₀₂ Rf	¹⁰³ ₁₀₃ Db	¹⁰⁴ ₁₀₄ Sg	¹⁰⁵ ₁₀₅ Bh	¹⁰⁶ ₁₀₆ Hs	¹⁰⁷ ₁₀₇ Mt	¹⁰⁸ ₁₀₈ Ds	¹⁰⁹ ₁₀₉ Rg	¹¹⁰ ₁₁₀ Cn	¹¹¹ ₁₁₁ Uut	¹¹² ₁₁₂ Fl	¹¹³ ₁₁₃ Uup	¹¹⁴ ₁₁₄ Lv	¹¹⁵ ₁₁₅ Uus	¹¹⁶ ₁₁₆ Uuo																													
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Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎

Matter:

Anything that has **mass** and **occupies space** ✎

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** ✎

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

Chemistry Ceoce Study Guide

**Kenneth Goldsby, Professor, Raymond
Chang, Dr.**

Chemistry Ceoce Study Guide:

Study Guide for Chemical Principles [by] Steven S. Zumdahl Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples

Chemistry and Student Study Guide and Study Guide, Fifth Edition and Smarthinking and Letter Zumdahl, 2002-01-01

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Study Guide for Introductory Chemistry Nivaldo Tro, Donna G. Friedman, 2017-07-09 This Study Guide was written specifically to assist students using the 6th edition of Introductory Chemistry It presents the major concepts theories and applications discussed in the text in a comprehensive and accessible manner for students It contains learning objectives chapter summaries and outlines as well as examples self tests and concept questions

Study Guide for Chemistry Theodore E. Brown, James C. Hill, 2014-02-07 This book assists students through the text material with chapter overviews learning objectives a review of key terms as well as self tests with answers and explanations This student guide also features MCAT practice questions

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Chemistry (Teacher Guide) Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master s Class Chemistry course by Master Books The teacher is one who guides students through the subject matter helps each student stay on schedule and be organized and is their source of accountability along the way With that in mind this guide provides additional help through the laboratory exercises as well as lessons quizzes and examinations that are provided along with the answers The lessons in this study emphasize working through procedures and problem solving by learning patterns The vocabulary is kept at the essential level Practice exercises are given with their answers so that the patterns can be used in problem solving These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college Guided labs are provided to enhance instruction of weekly lessons There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions It is important to see the hand of God and His principles and wisdom as it plays out in chemistry This course integrates what God has told us in the context of this study Features Each suggested weekly schedule has five easy to manage lessons that combine reading and worksheets Worksheets quizzes and tests are perforated and three hole punched

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Study Guide and Selected Solutions Manual for Introductory Chemistry Charles H. Corwin,2017-01-02

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Study Guide for Chemistry Karen C. Timberlake,2011-01-26 *Student Study Guide for Chemistry* Kenneth Goldsby, Professor,Raymond Chang, Dr.,2015-03-06 This valuable ancillary contains material to help the student practice problem solving skills For each section of a chapter the author provides study objectives and a summary of the corresponding text Following the summary are sample problems with detailed solutions Each chapter has true false questions and a self test with all answers provided at the end of the chapter

Student Study Guide to Chemistry for Changing Times John William Hill,1980-01-01

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