

General Chemistry II Chapter 17 Study Guide

Organic acids- carboxylic acids (R-COOH):

- Carboxylic acids (R-COOH): a weak organic acid, contains the carboxyl group (-COOH); never fully dissociates in water, an equilibrium process, ex. Formic acid, acetic acid, butyric acid, benzoic acid
 - R= rest of the molecule
- Strength of carboxylic acids: depends on the R group; more electronegative Cl in R group shifts more electron density towards the R group and helps stabilize the structure after the H⁺ ion is removed; thus, chloroacetic acid is the stronger acid in comparison to acetic acid (CH₃COOH)

Organic bases- amines:

- Amines: organic bases, contains an N atom with a lone pair (ex. -NH, -NH₂); ex. Methyl amine, dimethyl amine, trimethyl amine; are weak bases
 - Only N with lone pair can accept a proton
 - Equilibrium dissociation process
 - NH₄⁺ is an acid and can lose one H⁺ to be neutral

Organic acids and bases (things to keep in mind):

- Acid: H is the only acidic proton; ex. CH₃COOH becomes CH₃COO⁻
- Amine: the proton H⁺ is bonded to N via the lone pair of N; ex. CH₃NH₂ becomes CH₃NH₃⁺
- How about carbon?
 - Protons on C are not acidic; C cannot accept extra protons

Autolionization of water:

- Autolionization: in pure water, one water molecule can donate a proton to another water molecule



At 25 °C, $[\text{H}_3\text{O}^+] = [\text{OH}^-] = 1.0 \times 10^{-7} \text{ M}$

$$\begin{aligned}
 K_w &= [\text{H}_3\text{O}^+] [\text{OH}^-] = 1.0 \times 10^{-14} \\
 &\text{K}_w: \text{ion-product constant}
 \end{aligned}$$

Chemistry Chapter 17 Study Guide Answers

Susan McMurry

Chemistry Chapter 17 Study Guide Answers:

Study Guide and Solutions Manual Neil E. Schore, 2002-08-02 **Organic Chemistry Study Guide and Solutions Manual** Marc Loudon, Jim Parise, Lisa Bonner, 2024-08-09

The Study Guide and Solutions Manual offers several study tools for students. Study Guide Links are brief supplements to text material and provide hints for study shortcuts approaches to problem solving or additional explanations of topics that typically cause difficulty. Further Explorations are short discussions that provide additional depth on content in the text. Reaction Reviews present the reactions in the chapter in detailed outline form. This summary consolidates the reactions of the chapter in one place and presents not only the reactions themselves but also the essential features of their mechanisms and stereochemistry for cases in which these issues are discussed in the text. Solutions are provided for all of the over 1600 problems in the text. Many include worked out examples in which not only the answer is given but also the detailed approach to the solution is discussed.

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Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein,2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems

Student Study Guide Fundamentals of Chemistry by David E. Goldberg Margaret G. Kimble,1994

Study Guide/Selected Solutions Manual Julie R. Frentrup,2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book

Solutions Manual and Study Guide to Accompany Introduction to Organic Chemistry, 4th Ed Paul A. Bartlett,Judith G. Koch,1992

Study Guide with Solutions to Accompany Burgoyne, A Study Course in Organic Chemistry N. Sue Barnes,1979

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Evaluation,1978 **Study Guide and Solutions Manual for Fundamentals of Organic Chemistry** Susan McMurry,1994 Following a brief review of structure and bonding organic molecules and functional groups are presented as early as possible The text is organized primarily by functional group beginning with simple alkanes and moving toward more complex compounds Emphasis is placed on the fundamental mechanistic similarities of organic reactions McMurry s thorough revision continues to present the solid content necessary for this course without sacrifice of important subjects and

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[Organic Chemistry, Study Guide and Solutions Manual](#) T. W. Graham Solomons, Craig B. Fryhle, 2000 On the cover of this book is a Pacific yew tree found in the ancient forests of the Pacific Northwest. The bark of the Pacific yew tree produces Taxol, found to be a highly effective drug against ovarian and breast cancer. Taxol blocks mitosis during eukaryotic cell division. The supply of Taxol from the Pacific yew tree is vanishingly small; however, a single 100-year-old tree provides only about one dose of the drug, roughly 300 mg. For this reason, as well as the spectacular molecular architecture of Taxol, synthetic organic chemists fiercely undertook efforts to synthesize it. Five total syntheses of Taxol have thus far been reported. Now a combination of isolation of a related metabolite from European yew needles and synthesis of Taxol from that intermediate supply the clinical demand. This case clearly demonstrates the importance of synthesis and the use of organic

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