



Cobalt(III) chloride dihydrate  
( $\text{CoCl}_3 \cdot 2\text{H}_2\text{O}$ )

Ethanol

Mark

(a) An amount of solute is weighed out on an analytical balance and then

(b) A portion of the solvent is added to the volumetric flask.

(c) The mixture is swirled until all of the solute is dissolved.

(d) Additional solvent is added up to the mark on the volumetric flask.

# Chemistry Dimensions 1 Solutions

**Chemical Society (Great Britain)**

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General Chemistry Horace Grove Deming,1925      *Outlines of Theoretical Chemistry* Frederick Hutton Getman,1918  
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