

Nuclear Chemistry - Guided Practice Questions

Multiple Choice - Choose the one best answer and write its letter in the blank provided.

1) In balancing the nuclear reaction $^{238}_{\text{U}}\text{U} \rightarrow ^{206}_{\text{Pb}}\text{Pb} + \text{?}$, the identity of the missing element is $^{4}_{\text{He}}$. T

(A) Pu. (B) Cl. (C) Th. (D) He. (E) Pb.

2) The following reaction, $^{239}_{\text{Np}}\text{Pu} \rightarrow ^{235}_{\text{U}}\text{U} + \text{?}$, is an example of what form of radioactive decay?

(A) alpha. (B) positron. (C) gamma. (D) beta. (E) $\bar{\nu}$ -capture.

3) The missing product from this reaction is $^{235}_{\text{U}}\text{U} \rightarrow ^{141}_{\text{Ba}}\text{Ba} + \text{?}$

(A) $^{92}_{\text{Kr}}$. (B) $^{40}_{\text{Ar}}$. (C) $^{96}_{\text{Kr}}$. (D) $^{40}_{\text{Ar}}$. (E) $^{40}_{\text{Ar}}$.

4) What is the mass number of a neutron _____?

(A) 1. (B) 3. (C) 2. (D) 0. (E) 4.

5) In the nuclear decay reaction represented by $^{14}_{\text{N}}\text{N} \rightarrow ^{14}_{\text{N}}\text{N}$, the emitted particle is _____.

(A) beta particle. (B) neutron. (C) alpha particle. (D) positron. (E) proton.

6) The reaction shown below is responsible for creating C-14 in the atmosphere. What is the bombarding particle _____?



(A) electron. (B) proton. (C) positron. (D) alpha particle. (E) neutron.

7) $^{130}_{\text{I}}\text{I}$ has a half-life of 8.04 days. Assuming you start with a 2.50 mg sample of $^{130}_{\text{I}}$, how many mg will remain after 13.0 days _____?

(A) 0.440. (B) 0.422. (C) 0.298. (D) 0.193. (E) 0.499.

8) All atoms of a given element have the same:

(A) atomic mass. (B) mass number. (C) # of nucleons. (D) atomic #. (E) # of neutrons.

9) Atoms containing radioactive nuclei are called:

(A) radioisotopes. (B) radionuclides. (C) nuclides. (D) nucleons. (E) radioisotopes.

10) What happens to the mass number and the atomic number of an element when it undergoes beta decay?

(A) The mass number decreases by 4 and the atomic number decreases by 2. (B) The mass number does not change and the atomic number decreases by 2. (C) Neither the mass number nor the atomic number change. (D) The mass number increases by 2 and the atomic number increases by 1. (E) The mass number does not change and the atomic number increases by 1.

Chemistry Guided Practice Problems Answer Key

John S. Phillips, Cheryl Wistrom

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Organic Chemistry, a Guided Inquiry Andrei Straumanis,2004 Learning to Solve Complex Scientific Problems David H. Jonassen,2017-09-25 Problem solving is implicit in the very nature of all science and virtually all scientists are hired retained and rewarded for solving problems Although the need for skilled problem solvers has never been greater there is a growing disconnect between the need for problem solvers and the educational capacity to prepare them Learning to Solve Complex Scientific Problems is an immensely useful read offering the insights of cognitive scientists engineers and science educators who explain methods for helping students solve the complexities of everyday scientific problems Important features of this volume include discussions on how problems are represented by the problem solvers and how perception attention memory and various forms of reasoning impact the management of information and the search for solutions how academics have applied lessons from cognitive science to better prepare students to solve complex scientific problems gender issues in science and engineering classrooms and questions to guide future problem solving research The innovative methods explored in this practical volume will be of significant value to science and engineering educators and researchers as well as to instructional designers

Invitation to Organic Chemistry Alyn William Johnson,1999 Colorful graphics and 19 chapters featuring such learning aids as chemistry at work and conceptual problems characterize this large text on a large subject Cited by the American Association for the Advancement of Science for his pioneering work in the chemistry of ylides Johnson who spent most of his career at the U of North Dakota explores the smorgasbord of subject matter that is organic chemistry and new developments in the field Appends a summary of nomenclature spectra group assignments and values of selected important compounds The index is combined with a glossary Annotation copyrighted by Book News Inc Portland OR

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for Reaching struggling readers Working with English learners and their families Designing lessons with assessment in mind Sustaining a passion for teaching Covering everything needed for your first year and beyond this guide provides invaluable insights on succeeding with adolescent students

Chemistry John S. Phillips, Cheryl Wistrom, 2000

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emphasise the educational implications and applications of child development cognitive science learning and teaching Recurring themes throughout the text include ideas about education social and socio cultural aspects of education schools

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Australasian perspectives and applications using the work of Australasian researchers and teachers Numerous examples case studies guidelines and practical tips from experienced teachers are used in the text to explore the connections between knowledge understanding and practice

Mastery Learning in the Science Classroom Kelly Morgan, 2011 In these pages

Kelly Morgan presents a compelling case for implementing a mastery learning science classroom and then shows us how to do it Using research based student performance data Morgan compiles impressive statistics that support her assertion OC Mastery learning results in improved student learning and motivation OCO Showing challenges as well as benefits this text covers a step by step implementation from the traditional classroom to a mastery classroom

Understanding Chemistry

Fred M. Dewey, 1994

New Digital Technology in Education Wan Ng, 2015-04-25 This book addresses the issues

confronting educators in the integration of digital technologies into their teaching and their students learning Such issues

include a skepticism of the added value of technology to educational learning outcomes the perception of the requirement to keep up with the fast pace of technological innovation a lack of knowledge of affordable educational digital tools and a lack of understanding of pedagogical strategies to embrace digital technologies in their teaching This book presents theoretical

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data to existing scholarship in this field Examples of pedagogical possibilities that are both new and currently practised across a range of teaching contexts are featured

Action Research for Educators Daniel R. Tomal, 2010-03-15

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action research ethical and legal considerations methods for conducting both formal and informal action research data collection methods analysis and interpretation action planning and initiation and results evaluation The author includes

numerous examples strategies and illustrations that can be applied to elementary and secondary schools as well as university settings *Holt Chemistry* Ralph Thomas Myers,2004 The Elementary School Library Collection, Phases 1-2-3 ,1970

The Software Encyclopedia 2001 ,2001 **Science Books & Films** ,1989 Christian Home Educators' Curriculum Manual Cathy Duffy,1997-11 **CHEMISTRY** SILBERBERG,2003 **Tutoring** Ross Barclay Macdonald,1988 **Journal of the American Medical Association** American Medical Association,1910 Includes proceedings of the Association papers read at the annual sessions and list of current medical literature 1998 Educational Software Preview Guide Educational Software Evaluation Consortium,International Society for Technology in Education,1998 **Etude** ,1890 Includes music

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