

General Chemistry II Study Guide

Questions	Answers
At what speed do molecules move?	Molecules move at a variety of speeds depending on the amount of energy in them.
What could cause a molecule to escape the IMF in which it is held?	A large amount of energy would cause a molecule to escape the liquid that it is in.
What is vapor pressure?	The pressure (concentration) of gas molecules above the liquid.
Which molecules will have vapor pressure?	All molecules have some sort of vapor pressure
What is the relationship between IMFs and P _{vap} ?	As the strength of the IMF increases the P _{vap} decreases.
What is the relationship between temperature and P _{vap} ?	As temperature increases, P _{vap} also increases.
What is the definition of boiling?	When thermal energy is greater than the IMF
What is the real definition of boiling point?	When enough thermal energy has been added to give all the molecules the ability to escape.
What is the external pressure equal to?	1 atm
What does atm stand for?	atmosphere
If there are two jars, one with hydrogen bonding and one with dipole-dipole, which would you expect to have a higher vapor pressure?	Dipole-dipole
When a substance goes from a solid to	Sublimation.

Chemistry Ii Study Guide

**John Krenos, Joseph Potenza, Loretta
Jones, Lynn Koplitz, Thomas Spence**

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