

Naming Ionic Compounds Practice Worksheet

Name the following ionic compounds:

- 1) NH_4Cl _____
- 2) $\text{Fe}(\text{NO}_3)_3$ _____
- 3) TiBr_3 _____
- 4) Co_3P _____
- 5) SnSe_2 _____
- 6) GaAs _____
- 7) $\text{Pb}(\text{SO}_4)_2$ _____
- 8) $\text{Be}(\text{HCO}_3)_2$ _____
- 9) $\text{Mn}_2(\text{SO}_4)_3$ _____
- 10) $\text{Al}(\text{CN})_3$ _____

Write the formulas for the following compounds:

- 11) chromium (VI) phosphate _____
- 12) vanadium (IV) carbonate _____
- 13) tin (II) nitrite _____
- 14) cobalt (III) oxide _____
- 15) titanium (II) acetate _____
- 16) vanadium (V) sulfide _____
- 17) chromium (III) hydroxide _____
- 18) lithium iodide _____
- 19) lead (II) nitride _____
- 20) silver bromide _____

Lots of Ionic Naming Practice Problems

Name the following ionic compounds:

- 1) NaBr _____
- 2) $\text{Sc}(\text{OH})_3$ _____
- 3) $\text{V}_2(\text{SO}_4)_3$ _____
- 4) NH_4F _____
- 5) CaCO_3 _____
- 6) NiPO_4 _____
- 7) Li_2SO_3 _____
- 8) Zn_3P_2 _____
- 9) $\text{Sn}(\text{C}_2\text{H}_3\text{O}_2)_4$ _____
- 10) Cu_2O _____
- 11) Ag_3PO_4 _____
- 12) YClO_3 _____
- 13) SnS_7 _____
- 14) $\text{Ti}(\text{CN})_4$ _____
- 15) KMnO_4 _____
- 16) Pb_3N_2 _____
- 17) CoCO_3 _____
- 18) CdSO_3 _____
- 19) $\text{Cu}(\text{NO}_3)_2$ _____
- 20) $\text{Fe}(\text{HCO}_3)_2$ _____

Chemistry Naming Ions Study Guide Answer Key

Steven S. Zumdahl, Paul B. Kelter



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Table of Contents **Chemistry Naming Ions Study Guide Answer Key**

1. Understanding the eBook Chemistry Naming Ions Study Guide Answer Key
 - The Rise of Digital Reading Chemistry Naming Ions Study Guide Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Naming Ions Study Guide Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Naming Ions Study Guide Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Naming Ions Study Guide Answer Key
 - Personalized Recommendations
 - Chemistry Naming Ions Study Guide Answer Key User Reviews and Ratings
 - Chemistry Naming Ions Study Guide Answer Key and Bestseller Lists

5. Accessing Chemistry Naming Ions Study Guide Answer Key Free and Paid eBooks
 - Chemistry Naming Ions Study Guide Answer Key Public Domain eBooks
 - Chemistry Naming Ions Study Guide Answer Key eBook Subscription Services
 - Chemistry Naming Ions Study Guide Answer Key Budget-Friendly Options
6. Navigating Chemistry Naming Ions Study Guide Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Naming Ions Study Guide Answer Key Compatibility with Devices
 - Chemistry Naming Ions Study Guide Answer Key Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Chemistry Naming Ions Study Guide Answer Key
 - Highlighting and Note-Taking Chemistry Naming Ions Study Guide Answer Key
 - Interactive Elements Chemistry Naming Ions Study Guide Answer Key
8. Staying Engaged with Chemistry Naming Ions Study Guide Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Naming Ions Study Guide Answer Key
9. Balancing eBooks and Physical Books Chemistry Naming Ions Study Guide Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Naming Ions Study Guide Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Naming Ions Study Guide Answer Key
 - Setting Reading Goals Chemistry Naming Ions Study Guide Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Naming Ions Study Guide Answer Key
 - Fact-Checking eBook Content of Chemistry Naming Ions Study Guide Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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