



CHEMISTRY

CHEAT SHEET



STATES OF MATTER

STATES OF MATTER CHEAT SHEET

Matter		Physical and Chemical Changes	
Matter is anything that takes up space and has volume. It has mass and is made of atoms and molecules.		<p>Physical Change: A change in which the substance's physical properties change, but its chemical identity remains the same. Examples: melting, freezing, boiling, condensation, sublimation, deposition.</p> <p>Chemical Change: A change in which the substance's chemical identity changes. Examples: burning, rusting, digestion, photosynthesis.</p>	
Phase	Description	Arrangement of Particles	Examples
SOLID	<ul style="list-style-type: none"> Lowest energy state of matter. Particles are closely packed in a regular, repeating pattern. Highly resistant to shape and volume changes. 		
LIQUID	<ul style="list-style-type: none"> Intermediate state. Particles are close together but can move past each other. Definite volume but no definite shape. 		
GAS	<ul style="list-style-type: none"> Highest energy state of matter. Particles are far apart and move rapidly in all directions. No definite shape or volume. 		
PLASMA	<ul style="list-style-type: none"> Highest energy state of matter. Consists of free electrons and positive ions. 		
Relative Energy Levels: The energy levels of matter are ordered from lowest to highest as follows: Solid < Liquid < Gas < Plasma.		Relative Energy Levels: Solid Liquid Gas Plasma	

Chemistry States Of Matter Study Guide

College Entrance Examination Board



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Resources in Education, 1984 *Study Guide with Answers to Selected Problems* H. Stephen Stoker, Danny V. White, 2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises

Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter

CLEP Official Study Guide The College Board, 2005-08-20 The CLEP Official Study Guide 17th Edition is the only complete source of information about the CLEP exams You get Descriptions of the exams to help students decide which exams to take Sample questions and answers for all 35 exams Valuable test taking tips and information on obtaining credit through CLEP With more than 6 million CLEP exams taken since 1967 the College Level Examination Program is a credit by examination program that has proven to be effective for students advisors and faculty Students can demonstrate their comprehensive subject knowledge gain credit and move on to more advanced topics Advisors can help students plan their undergraduate years by helping them shorten the time needed to complete their degree Faculty can use their undergraduate introductory courses more effectively There are exams in 35 introductory college level subjects and are administered on

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