

CHEMISTRY

CONVERSION CHART

PREFIX	ABBR.	EXAMPLES USING G AS BASE UNIT	CONVERSION NEAREST NEIGHBOR
tera-	T	1,000,000,000,000 g = t Tg	1,000 Gg = 1 Tg
giga-	G	1,000,000,000 g = 10 ⁹ g	1,000 Mg = 10 ⁹ g
mega-	M	1,000,000 g = 1 Mg	1,000 kg = 1 Mg
kilo-	k	1,000 g = 1 kg	100 Dg = 1 kg
deca-	D	10 g = 1 Dg	
Base unit			
deci-	d	10 dg = 1 g	
centi-	c	100 cg = 1 g	10 cg = 1 dg
milli-	m	1,000 mg = 1 g	10 mg = 1 cg
micro-	μ	1,000,000 μg = 1 g	1,000 μg = 1 mg
nano-	n	1,000,000,000 ng = 1 g	1,000 ng = 1 μg
pico-	p	1,000,000,000,000 pg = 1 g	1,000 pg = 1 ng
femto-	f	1,000,000,000,000,000 = 1 g	1,000 fg = 1 pg

Conversion Tables For Chemistry

François Cardarelli



Conversion Tables For Chemistry:

Scientific Unit Conversion Francois Cardarelli, 2012-12-06 1 2 How to Use This Book Chapter 2 contains a brief history of the metric system including the organization and a complete description of SI Units Systeme International d Unites Chapter 3 gives a detailed description of a considerable number of other systems of measurement This includes several alternative modern systems of measurement some of which are still in widespread use e g imperial US cgs MTS FPS Finally there is a description of systems used in antiquity e g ancient Chinese Indian Egyptian Persian Hebrew Greek Roman Arabic as well as older national or regional systems e g French Italian German Japanese Chapter 4 which forms the most important part of the book consists of an exhaustive set of conversion tables This chapter covers the units in alphabetical order Each unit is fully described as follows name symbol s physical quantity dimension conversion factor notes and definitions The section covers some 2000 units each with a precise conversion factor Chapter 5 enables a unit to be identified from its area of application For this purpose units are classed in groups It contains thirty five conversion tables ranging from mass to nuclear quantities In order to facilitate use of this manual several supplementary sections have been added to aid the researcher These include tables of fundamental mathematical and physical constants to allow very precise calculation of conversions These form the sixth chapter of the book

Chemical Conversion Tables F. B. Dancy, Herbert Bemerton Battle, 1885 Catalog of Books and Reports in the Bureau of Mines Technical Library, Pittsburgh, Pa United States. Bureau of Mines. Technical Library, Pittsburgh, 1968 CRC Handbook of Radioanalytical Chemistry Volume 1 Juraj Tolgyessy, Erno Bujdosó, 2024-11-01 Radioanalytical methods have become among the most important means for elemental analysis and the determination of chemical species Their extreme sensitivity has made them indispensable in a wide range of applications including mineral analysis medical and biophysical work criminology history archaeology and space research This handbook combines theoretical and practical radioanalytical work covering the entire field of radioanalytical chemistry Topics discussed include analysis by activation and nuclear reactions isotope dilution analysis radioreagent methods analysis by absorption and the scattering of radiation The handbook is extremely useful to scientists conducting applied and basic research in subjects related to analytical measurements engineers designing control facilities and equipment and professors and students working with analytical chemistry radiochemistry radioanalytical chemistry environmental chemistry biology and physics

Chemical Conversion Tables F B Dancy, Herbert B Battle, 2025-05-22 Chemical Conversion Tables 1885 is a comprehensive reference work designed for chemists pharmacists and students Authored by F B Dancy and Herbert B Battle this book provides extensive tables for converting weights measures and chemical formulas It is an invaluable tool for accurately performing calculations and conversions essential in chemical analysis and experimentation This historical text offers insights into the methodologies and standards of late 19th century chemistry making it a valuable resource for historians of science and anyone interested in the evolution of chemical practices Its detailed tables and

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