

1

SOME BASIC CONCEPTS OF CHEMISTRY

Weightage 8 marks

1 mark - 1Q (Q No. 1)

2 mark - 1Q (Q No. 11)

5 mark - 1Q (Q No. 27)

ONE MARK QUESTIONS

1. Following medicines are used in the treatment of which disease?

- (a) Cisplatin (PUB)
(b) Taxol (EQ)
(c) Drug AZT (Azidothymidine) (EQ)

Ans: (a) Treatment of Cancer
(b) Treatment of Cancer
(c) Treatment of AIDS patients.

2. Give the S.I. unit of

- (i) Density (PUB, May-17, Feb -17,19, 20)
(ii) Temperature (May 15)
(iii) Amount of substance (Feb 18)
(iv) Electric current (Feb 19)
(v) Luminous intensity
(vi) Volume (Feb 20)

Ans: (i) kg/m^3 (ii) Kelvin (iii) Mole
(iv) Ampere (v) Candela (vi) m^3

3. Express the following in scientific notation

- (i) 0.00016 (Feb 18)

Ans: 1.6×10^{-4}

- (ii) 0.00085 (May 15)

Ans: 8.5×10^{-4}

- (iii) 0.00053 (Feb 16)

Ans: 5.3×10^{-4}

- (iv) 8008 (EQ)

Ans: 8.008×10^3

- (v) 0.001023 (PUB)

Ans: 1.023×10^{-3}

- (vi) 232.508 (Feb 19)

Ans: 2.32508×10^2

4. How many significant figures are there in

- (i) 0.00425 (Feb 17)
(ii) 20.0
(iii) 6.022×10^{23}
(iv) 0.2500 (PUB)
(v) 3.10087
(vi) 8.2015

Ans: (i) 3 (ii) 3 (iii) 4 (iv) 4 (v) 6 (vi) 5

5. Give the relation between

- (a) Degree Fahrenheit and degree Celsius. (Feb 17)
(b) Degree Celsius and Kelvin (EQ)

Ans: (a) $^{\circ}\text{F} = 9/5 \times (^{\circ}\text{C}) + 32$
(b) $\text{K} = ^{\circ}\text{C} + 273$

6. State law of Conservation of Mass (EQ)

Ans: It states that "matter can neither be created nor destroyed".

7. State law of Definite Proportions (PUB)

Ans: The law states that a given compound always contains exactly the same proportion of elements by weight.

8. State law of 'Multiple Proportions'. (EQ)

Ans: When two elements can combine to form more than one compound, the masses of one element that combine with a fixed mass of the other element, are in the ratio of small whole numbers.

9. State 'Gay Lussac's Law' of gaseous volumes. (EQ)

Ans: The volume of gases taking part in a chemical reaction show simple whole number ratio to one another when all the volumes are measured at the same temperature and pressure.

10. State Avogadro's law. (May14, Feb-17,20)

Ans: It states that equal volumes of all gases contain equal number of molecules at same temperature and pressure.

Chemistry Study Guide Questions And Concepts

Matthew Bowling



Chemistry Study Guide Questions And Concepts:

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