

1

SOME BASIC CONCEPTS OF CHEMISTRY

Weightage 8 marks

1 mark - 1Q (Q No. 1)

2 mark - 1Q (Q No. 11)

5 mark - 1Q (Q No. 27)

ONE MARK QUESTIONS

1. Following medicines are used in the treatment of which disease?

(a) Cisplatin (PUB)
(b) Taxol (EQ)
(c) Drug AZT (Azidothymidine) (EQ)

Ans: (a) Treatment of Cancer
(b) Treatment of Cancer
(c) Treatment of AIDS patients.

2. Give the S.I. unit of

(i) Density (PUB, May-17, Feb-17, 19, 20)
(ii) Temperature (May 15)
(iii) Amount of substance (Feb 18)
(vi) Electric current (Feb 19)
(v) Luminous intensity
(vi) Volume (Feb 20)

Ans: (i) kg/m³ (ii) Kelvin (iii) Mole
(iv) Ampere (v) Candela (vi) m³

3. Express the following in scientific notation

(i) 0.00016 (Feb 18)

Ans: 1.6×10^{-4}

(ii) 0.00085 (May 15)

Ans: 8.5×10^{-4}

(iii) 0.00053 (Feb 16)

Ans: 5.3×10^{-4}

(iv) 8008 (EQ)

Ans: 8.008×10^3

(v) 0.001023 (PUB)

Ans: 1.023×10^{-3}

(vi) 232.508 (Feb 19)

Ans: 2.32508×10^2

4. How many significant figures are there in
(i) 0.00425 (Feb 17)
(ii) 20.0
(iii) 6.022×10^{23}
(iv) 0.2500 (PUB)
(v) 3.10087
(vi) 8.2015

Ans: (i) 3 (ii) 3 (iii) 4 (iv) 4 (v) 6 (vi) 5

5. Give the relation between

(a) Degree Fahrenheit and degree Celsius. (Feb 17)
(b) Degree Celsius and Kelvin (EQ)

Ans: (a) ${}^{\circ}\text{F} = \frac{9}{5} \times ({}^{\circ}\text{C}) + 32$
(b) $\text{K} = {}^{\circ}\text{C} + 273$

6. State law of Conservation of Mass (EQ)

Ans: It states that "matter can neither be created nor destroyed".

7. State law of Definite Proportions (PUB)

Ans: The law states that a given compound always contains exactly the same proportion of elements by weight.

8. State law of "Multiple Proportions". (EQ)

Ans: When two elements can combine to form more than one compound, the masses of one element that combine with a fixed mass of the other element, are in the ratio of small whole numbers.

9. State "Gay Lussac's Law" of gaseous volumes. (EQ)

Ans: The volume of gases taking part in a chemical reaction show simple whole number ratio to one another when all the volumes are measured at the same temperature and pressure.

10. State Avogadro's law. (May 14, Feb-17, 20)

Ans: It states that equal volumes of all gases contain equal number of molecules at same temperature and pressure.

Chemistry Study Guide Questions And Concepts

Matthew Bowling

Chemistry Study Guide Questions And Concepts:

Basic Concepts of Chemistry, Study Guide Leo J. Malone, 1989-02-21 This Third Edition revised to provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes many illustrative worked examples

Robinson Chemistry Study Guide Robinson, 1992

Study Guide with Selected Solutions Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

[Chemistry - Study Guide to Accompany Moore/Davies/Collins](#) Clare T. Furse, 1978

Chemistry Richard Post, Chad Snyder, Clifford C. Houk, 2020-09-16

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A

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